Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2024-25 **Semester: Even**

Name of Asstt./Ass. Prof /Professor: Dr. Ravinder Singh Class: B.Sc. 2nd /4th /6th Semester (Chem. Hons.)

16 th January 2025 to 31 st	May 2025 [B.Sc. 2 nd Semester Single Major (Chemistry)]
Week 1	Chemistry of Halogenated Hydrocarbons
30 th January 1 st	Alkyl Halides: Methods of preparation and properties
February 1	Aikyi Haildes: Methods of preparation and properties
2 th February	SUNDAY
-	
Week 2	Chemistry of Halogenated Hydrocarbons
3 th February – 8 th	Alkyl Halides: Nnucleophilic substitution reactions SN1, SN2 and SNi mechanisms with
February	stereochemical aspects and factors affecting the rate of SN reactions. Elimination reactions-
	E1, E2 and E1cB mechanism, nucleophilic substitution vs elimination.
9 th February	SUNDAY
Week 3	Chemistry of Halogenated Hydrocarbons
10 th February – 15 th	Aryl Halides: Preparations (including preparation from diazonium salts) and
February	properties, nucleophilic aromatic substitution, SNAr, benzyne mechanism.
	Relative reactivity of alkyl, allyl, benzyl, vinyl and aryl halides towards
	nucleophilic substitution reactions. Organometallic compounds of Mg (Grignard reagent)- use
	in synthesis of organic compounds.
16 th February	SUNDAY
Week 4	Revision, Assignment and Test
7 th February–22 nd	
February	
23 rd February	SUNDAY
Week 5	Acids and Bases: Arrhenius, Bronsted-Lowry, Lux-flood, solvent system and Lewis concept
24 th February–2 nd	of acids and bases,
March	,
3 rd March	SUNDAY
Week 6	Acids and Bases: Relative strength of acids and bases, levelling effect, classification of acids
4 th March–9 th March	and bases as hard and soft. Pearson's HSAB concept, applications of HSAB principle (acid-
	base strength and hardness and softness). Symbiosis, theoretical basis of hardness and softness,
	electronegativity.
10 th March	SUNDAY
Week 7	Non-Aqueous Solvents: Physical properties of a solvent, types of solvents and their general
11 th March– 16 th March	characteristics, reactions in non-aqueous solvents with reference to liquid HF, NH ³ and liquid
	SO^2 .
17 th March	SUNDAY
Week 8	Revision, Assignment and Test
18 th March – 23 rd	
March	
24 th March	SUNDAY
Week 9	Chemistry of p- Block Elements (groups 13-17): Inert pair effect, diagonal
25 th March – 30 th	relationship, general characteristics of groups 13-17 elements.
25 th March – 30 th March 31 th March	relationship, general characteristics of groups 13-17 elements. SUNDAY

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2024-25 **Semester: Even**

Name of Asstt./Ass. Prof /Professor: Dr. Ravinder Singh Class: B.Sc. 2nd /4th /6th Semester (Chem. Hons.)

Week 10	Chemistry of p- Block Elements (groups 13-17): Preparations, properties, structure and
1 st April – 6 th April	uses of compounds like oxides, oxy-acids (comparison of acidic strength) and halides of
	groups 13-17, hydrides of boron - diborane, borazine (preparation and uses), boric acids and
	borax.
7 th April	SUNDAY
Week 11	
8 th April – 13 th April	Chemistry of p- Block Elements (groups 13-17): Chemistry of fullerenes, silicates and type
8 April – 13 April	of silicates on structural basis, silicones, phosphonitrilic halides $\{(PNCl^2)n \text{ where } n = 3 \text{ and } l$
4.4fb A •1	4}, basic properties of halogens, interhalogens, pseudohalogen and polyhalides.
14 th April	SUNDAY
Week 12	Revision, Assignment and Test
15 th April – 20 th April	
21st April	SUNDAY
Week 13	Thermodynamics-II
22 nd April – 27 th April	Second Law: Concept of entropy, thermodynamic scale of temperature, statement of second
	law of thermodynamics. Carnot's cycles and its efficiency, Carnot's theorem, calculation of
	entropy change for reversible and irreversible processes.
28 th April	SUNDAY
Week 14	Thermodynamics-II
29 th April – 4 th May	Third Law: Statement of third law, concept of residual entropy, calculation of
	absolute entropy of molecules. Free Energy Functions: Gibbs and Helmholtz energy, variation
	of S, G, A with T, V, P, free energy change and spontaneity. Gibbs-Helmholtz equation,
	Maxwell relations, thermodynamic equation of state.
5 th May	SUNDAY
Week 15	Revision, Assignment and Test
6 th May – 11 th May	
21st April	SUNDAY
Week 16	Revision, Assignment and Test
22 nd May – 27 th May	
28 th April	SUNDAY
Week 17	Revision, Assignment and Test
29 th May – 4 th May	
5 th May	SUNDAY
Week 18	Revision, Assignment and Test
6 th May – 11 th May	
12 th May	SUNDAY
Week 19	Revision, Assignment and Test
13 th May – 18 th May	CUNDAY
19 th May	SUNDAY Revision Aggingment and Test
Week 20	Revision, Assignment and Test
20 th May – 25 th May	CINDAY
26 th May Week 21	SUNDAY Povision Assignment and Test
27 th May – 31 st May	Revision, Assignment and Test
41 May - 31 May	

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2024-25 **Semester: Even**

Name of Asstt./Ass. Prof /Professor: Dr. Ravinder Singh Class: B.Sc. 2nd /4th /6th Semester (Chem. Hons.)

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2024-25 **Semester: Even**

Name of Asstt./Ass. Prof /Professor: Dr. Ravinder Singh Class: B.Sc. 2nd /4th /6th Semester (Chem. Hons.)

Name of Subject: Genera	
7 th January 2025 to 11 th I	
Week 1	Aldehydes and Ketones
7th January – 11th	Nomenclature and structure of the carbonyl group. Synthesis of aldehydes and ketones with
January	particular reference to the synthesis of aldehydes from acid chlorides, advantage of oxidation
	of alcohols with chromium trioxide (Sarett reagent) pyridinium chlorochromate (PCC) and
	pyridinium dichromate., * synthesis of aldehydes and ketones using 1,3-dithianes,
	*Gatterman aldehyde synthesis, *Gatterman Koch reaction, *synthesis of ketones from
	nitriles and from carboxylic acids.
12 nd January	SUNDAY
Week 2	Aldehydes and Ketones
13 th January – 18 th	Physical properties. Comparison of reactivities of aldehydes and ketones. Mechanism of
January	nucleophilic additions to carbonyl group with particular emphasis on benzoin, aldol, Perkin and
19 th January	Knoevenagel condensations. Condensation with ammonia and its derivatives. SUNDAY
Week 3	
20 th January - 25 th	Aldehydes and Ketones Wittig reaction, Mannigh reaction, *Michael reaction, * Use of coatele as protecting group.
January January	Wittig reaction. Mannich reaction, *Michael reaction. * Use of acetals as protecting group.
Januar y	Oxidation of aldehydes, Baeyer-Villiger oxidation of ketones, Cannizzaro reaction. MPV,
	Clemmensen, Wolff-Kishner, LiAlH4 and NaBH4 reductions. * Halogenation of enolizable
	ketones. *An introduction to α , β -unsaturated aldehydes and ketones.
26 th January	SUNDAY
Week 4	Revision, Assignment and Test
27 th January 1 st	
February	CENTRALE
2 th February	SUNDAY
Week 5 3 th February – 8 th	Amines
•	Structure and nomenclatu re of amines, physical properties. Stereochemistry of amines.
February	Separation of a mixture of primary, secondary and tertiary amines. Structural features affecting
. 4	basicity of amines. *Amine salts as phase-transfer catalysts.
9 th February	SUNDAY
Week 6	Amines
10 th February – 15 th	Prepa ration of alkyl and aryl amines (reduction of nitro compounds, nitriles, reductive
T-1	
February	amination of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann
February	amination of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann bromamide reaction. Electrophilic aromatic substitution in aryl amines, reactions of amines
·	amination of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann bromamide reaction. Electrophilic aromatic substitution in aryl amines, reactions of amines with nitrous acid.
February 16 th February	amination of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann bromamide reaction. Electrophilic aromatic substitution in aryl amines, reactions of amines
·	amination of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann bromamide reaction. Electrophilic aromatic substitution in aryl amines, reactions of amines with nitrous acid. SUNDAY
16 th February	amination of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann bromamide reaction. Electrophilic aromatic substitution in aryl amines, reactions of amines with nitrous acid. SUNDAY Nitro Compounds
16 th February Week 7	amination of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann bromamide reaction. Electrophilic aromatic substitution in aryl amines, reactions of amines with nitrous acid. SUNDAY Nitro Compounds Preparation of nitro alkanes and nitro arenes and their chemical reactions. Mechanism of
16 th February Week 7 7 th February–22 nd	amination of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann bromamide reaction. Electrophilic aromatic substitution in aryl amines, reactions of amines with nitrous acid. SUNDAY Nitro Compounds Preparation of nitro alkanes and nitro arenes and their chemical reactions. Mechanism of electrophilic substitution reactions in nitro arenes and their reductions in acidic, neutral and
16 th February Week 7 7 th February–22 nd February	amination of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann bromamide reaction. Electrophilic aromatic substitution in aryl amines, reactions of amines with nitrous acid. SUNDAY Nitro Compounds Preparation of nitro alkanes and nitro arenes and their chemical reactions. Mechanism of electrophilic substitution reactions in nitro arenes and their reductions in acidic, neutral and alkaline medium. *Picric acid. Halonitroarenes: reactivity
16 th February Week 7 7 th February–22 nd February	amination of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann bromamide reaction. Electrophilic aromatic substitution in aryl amines, reactions of amines with nitrous acid. SUNDAY Nitro Compounds Preparation of nitro alkanes and nitro arenes and their chemical reactions. Mechanism of electrophilic substitution reactions in nitro arenes and their reductions in acidic, neutral and alkaline medium. *Picric acid. Halonitroarenes: reactivity SUNDAY
16 th February Week 7 7 th February–22 nd February	amination of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann bromamide reaction. Electrophilic aromatic substitution in aryl amines, reactions of amines with nitrous acid. SUNDAY Nitro Compounds Preparation of nitro alkanes and nitro arenes and their chemical reactions. Mechanism of electrophilic substitution reactions in nitro arenes and their reductions in acidic, neutral and alkaline medium. *Picric acid. Halonitroarenes: reactivity

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2024-25 **Semester: Even**

Name of Asstt./Ass. Prof /Professor: Dr. Ravinder Singh Class: B.Sc. 2nd /4th /6th Semester (Chem. Hons.)

3 rd March	SUNDAY
Week 9	Nitro Compounds
4th March-9th March	chemical reactions. Mechanism of electrophilic substitution reactions in nitro arenes and their
	reductions in acidic, neutral and alkaline medium. *Picric acid. Halonitroarenes: reactivity
10 th March	SUNDAY
Week 10	Diazonium Salts
11 th March– 16 th March	Mechanism of diazotisation, structure of benzene diazonium chloride, Replacement of diazo
	group by H, OH, F, Cl, Br, I, NO2 and CN groups, reduction of diazonium salts to hyrazines,
	coupling reaction and its synthetic application. * Preparation and reactions of cyanides, and
	isocyanides, urea and diazomethane.
17 th March	SUNDAY
Week 12	Infrared (IR) absorption spectroscopy
18 th March – 23 rd	Molecular vibrations, Hooke's law, selection rules, intensity and position of IR bands,
March	Workedian violations, frooke's law, selection fules, intensity and position of the bands,
24 th March	SUNDAY
Week 14	Infrared (IR) absorption spectroscopy
25 th March - 30 th	measurement of IR spectrum, fingerprint region,
March	
31 th March	SUNDAY
Week 15	Infrared (IR) absorption spectroscopy
1 st April – 6 th April	characteristic absorptions of various functional groups and interpretation of IR spectra of simple
	organic compounds.
7 th April	SUNDAY
Week 16	Infrared (IR) absorption spectroscopy
8 th April – 13 th April	*Hydrocarbons (saturated and unsaturated), hydroxy compounds, aldehydes, ketones,
	esters, anhydrides, amides, amines and nitrocompounds.
14 th April	SUNDAY
Week 17	Infrared (IR) absorption spectroscopy
15 th April – 20 th April	Applications of IR spectroscopy in structure elucidation of simple organic compounds.
21st April	SUNDAY
Week 18	Revision, Assignment and test
22 nd April – 27 th April	
28 th April	SUNDAY
Week 19	Revision, Assignment and test
29 th April – 4 th May	
5 th May	SUNDAY
Week 20	Revision, Assignment and test
6 th May – 11 th May	

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2024-25 **Semester: Even**

Name of Asstt./Ass. Prof /Professor: Dr. Ravinder Singh Class: B.Sc. 2nd /4th /6th Semester (Chem. Hons.)

7 th January 2025 to 11 th N	May 2025 [B.Sc. 6 th Semester Chemistry Hons.]
Week 1	Topics in Biological Chemistry: Introduction to enzymes, nomenclature, characteristics.
^{7th} January – 11 th	Practicals:
January	
12 nd January	SUNDAY
Week 2	Topics in Biological Chemistry: General picture of mechanism of enzymes action, co-
13 th January – 18 th	enzymes: co-enzymes derived from niacin and thiamine, lipoic acid, co-enzyme- A.
January	Practicals:
19 th January	SUNDAY
Week 3	Topics in Biological Chemistry: Energy production in biological system, glycolysis,
20 th January - 25 th	tricarboxylic acid cycle
January	Practicals:
26 th January	SUNDAY
Week 4	Revision, Assignment and test
27 th January 1 st	Practicals:
February	
2 th February	SUNDAY
Week 5	Fermentation
3 th February – 8 th	Anaerobic and aerobic fermentation, production of alcohol, citric acid and lactic acid
February	Practicals:
9 th February	SUNDAY
Week 6	Fats, Oil and Detergents:
10 th February – 15 th	Occurrence, chemical composition and importance, hydrogenated oils, Rancidity, acid
February	value, sponification and iodine numbers.
	Practicals:
16 th February	SUNDAY
Week 7	Fats, Oil and Detergents:
7 th February–22 nd	Difference between toilet and washing soaps, comparison o f soap and detergents,
February	classification and principle of cleansing action of detergents.
	Practicals:
23 rd February	SUNDAY
Week 8	Revision, Assignment and test
24 th February–2 nd	Practicals:
March	
3 rd March	SUNDAY
Week 9	Drugs-I
4 th March–9 th March	Introduction, relation of chemical structure and physiological activity with suitable
	examples.
	Practicals:
10 th March	SUNDAY

Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2024-25 **Semester: Even**

Name of Asstt./Ass. Prof /Professor: Dr. Ravinder Singh Class: B.Sc. 2nd /4th /6th Semester (Chem. Hons.)

Week 10	Drugs-I
11 th March–16 th March	Mechanism of chemotherapeutic action. Nomenclature of organic chemical systems,
	stereochemical notations.
	General aspects, preparation and uses of the following drugs:
	(i) Analgesics and antipyretics : paracetamol, Aspirin
	(i) Anti-inflammatory: Ibuprofen
	(iii) Antiseptics and disinfectants Chloro cresol, povidone – Iodine
	Practicals:
17 th March	SUNDAY
Week 12	Drugs-II
18 th March – 23 rd	General aspects, preparation and uses of the following drugs:
March	(i) Sulpha: Sulphacetamide
	(ii) Local anaesthetics : Benzocaine
	Practicals:
24 th March	SUNDAY
Week 14 25 th March – 30 th	Drugs-II
March	General aspects, preparation and uses of the following drugs:
March	(iii) Anti amoebic : Metronidazole
	(iv) Antimalarials : Chloroquinine
	Practicals:
31 th March	SUNDAY
Week 15	Revision, Assignment and test
1 st April – 6 th April	Practicals:
7 th April	SUNDAY
Week 16	Drugs-II
8 th April – 13 th April	General aspects, preparation and uses of the following drugs:
	(v) Antihistamines : Chlorophenizamine Maleate
	(vi) Antifungal Undecylenic acid
	Practicals:
14 th April	SUNDAY
Week 17	Drugs-II
15 th April – 20 th April	General aspects, preparation and uses of the following drugs:
	(vii) Insect repellants: Dibutyl phthalate
	(viii) Antibiotics Chloroamphenicol
	Practicals:
21st April	SUNDAY
Week 18	Revision, Assignment and test
22 nd April – 27 th April	Practicals:
28th April	SUNDAY
Week 19	Revision, Assignment and test
29th April – 4th May	Practicals:
5 th May	SUNDAY
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Name of College: Pt. Neki Ram Sharma Government College, Rohtak Academic Session 2024-25 **Semester: Even**

Name of Asstt./Ass. Prof /Professor: Dr. Ravinder Singh Class: B.Sc. 2nd /4th /6th Semester (Chem. Hons.)

Week 20	Revision, Assignment and test
6 th May – 11 th May	Practicals:
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Name of College: Pt. N.R.S. Government College, Rohtak

Academic Session:2024- 25 Semester: Even Name of Asstt./Ass.Prof: Dr. Ruman Rani

& Kiran Bala

Class: B.Sc.4th sem (N.M, Med.) Sec A & B Name of Subject: Physical Chemistry

*** 1 4	Winter Break
Week 1	
1 st –6 th Jan	
W 1.0	Thermodynamics-III Second law of thermodynamics, need for the law, different statements of the
Week 2	law ,Carnot's cycles and its efficiency, Carnot's theorm, Thermodynamics scale of temperature.
7 st –11 th Jan	
	SUNDAY
12 th Jan	
Week 3	Concept of entropy – entropy as a state function, entropy as a function of V & T, entropy as a function of P & T, entropy change in physical change,
VV COIL C	sundan of the try officers of the state of t
13st-18th Jan	
	SUNDAY
19 th Jan	
XX71 - 4	entropy as a criteria of spontaneity and equilibrium. Entropy change in ideal gases and mixing of
Week 4	gases.
20st–25thJan	
	SUNDAY
26 th Jan	
Week 5	Thermodynamics-IV Third law of thermodynamics: Nernst heat theorem,
,,, 5512 5	Thermosty manner in the control of t
27 st –1 th Feb	
	SUNDAY
2 nd Feb	
Week 6	statement of concept of residual entropy, evaluation of absolute entropy from heat capacity data.
vv eek o	
3 rd -8 th Feb	
	SUNDAY
9 th Feb	
	Gibbs and Helmholtz functions; Gibbs function (G) and Helmholtz function (A) as thermodynamic
Week 7	quantities
10 st –15 th Feb	
16 th Feb	SUNDAY
	A & G as criteria for thermodynamic equilibrium and spontaneity, their advantage over entropy
Week 8	change. Variation of G and A with P, V and T.
17 st –22 th Feb	
23 th Feb	SUNDAY
	Electrochemistry-III Electrolytic and Galvanic cells - reversible & Irreversible cells
Week 9	conventional representation of electrochemical cells. EMF of cell and its measurement, Wes to

	standard cell,
24 st –1 st March	Standard Cerr,
24°-1° Warth	SUNDAY
and v 4 1-	SUNDAY
2 nd March	
W I- 10	Activity and activity coefficients. Calculation of thermodynamic quantities of cell reaction (G,
Week 10	H & K).
3 rd –8 th March	
Week 11	Holi Break
9 th -16 th March	
	Types of reversible electrodes – metal- metal ion gas electrode, metal –insoluble salt- anion and
Week 12	redox electrodes. Electrode reactions
VVCCR 12	Found of the control
17 th –22 th March	
	SUNDAY
23 th March	
1,141,011	Nernst equations, derivation of cell EMF and single electrode potential.
Week 13	Therist equations, derivation of een Elvir and single electrode potential.
WCCK 13	
24 th –29 th March	
30 th March	SUNDAY
O O IVILLI CII	Standard Hydrogen electrode, reference electrodes, standard electrodes potential, sign conventions,
Week14	electrochemical se ries and its applications.
VVCCRI	Tr the transfer of the transfe
31 st –5 th April	
6 th April	SUNDAY
o riprii	
Week 15	Electrochemistry-IV Concentration cells with and without transference, liquid junction potential,
7 th –12 th April	
-	application of EMF measurement i.e. valency of ions, solubility product activity coefficient,
13 th April	SUNDAY
***	potentiometric titration (acid- base and redox). Determination of pH using Hydrogen electrode,
Week 16	Quinhydrone electrode and glass electrode by potentiometric methods.
1 4th 10th 4 9	
14 th -19 th April	CUINDAN
20 th April	SUNDAY
	Revision
Week 17	
21st–26 th April	
27 th April	SUNDAY
	Revision
Week 18	
28th-3rd May	

Name of College: Pt. N.R.S. Government College, Rohtak

Academic Session:2024- 25 Semester: Even Name of Asstt./Ass.Prof: Nidhi Mann, Anil

& Rinku

Class: B.Sc.2nd sem (DSC- II) Sec A, B &

C

Name of Subject: Chemistry

1 table of Sub	ject: Chemistry
	Init I
Week 1	Unit—I Non-aqueous Solvents Physical properties of a solvent, types of solvents and their general characteristics, solvent system concept
1 st –8 th Feb	
	SUNDAY
9 th Feb	
Week 2	reactions in non-aqueous solvents with reference to liquid NH3 and liquid SO2. Hard and soft acids and bases (HSAB concept), applications of HSAB principle
10st-15thFeb	
16 th Feb	SUNDAY
10 100	nature of bonding in noble gas compounds (valence bond treatment and MO treatment for XeF2 and
Week 3	XeF4), molecular shapes of noble gas compounds (VSEPR theory). Noble Gases Occurrence and uses
17 st –22 th Feb	
23 th Feb	SUNDAY
Week 4	, rationalization of inertness of noble gases, clathrates, preparation and properties, chemical properties of the noble gases, chemistry of xenon: structure and bonding in xenon fluorides, oxides and oxyfluorides (XeF2, XeF4, XeF6, XeO3, XeO4, XeOF2, XeO2F2, XeOF4, XeF5 + , XeF5 -)
24 st –1 st March	
	SUNDAY
2 nd March	
Week 5	Unit-II Thermodynamics Brief discussion upto first law of thermodynamics, heat capacity, heat capacities at constant volume and pressure and their relationship.
3 rd –8 th March	constant volume and pressure and then relationship.
Week 6	Holi Break
9 th -16 th March	
Week 7	Joule's law, Joule-Thomson coefficient for ideal gases and real gases and inversion temperature entropy as a function of V & T, entropy as a function of P & T, entropy as a function of P & V.
17 th –22 th March	
17 22 March	SUNDAY
23 th March	
25 Maich	calculation of work and heat, dU & dH for the expansion of ideal gases and real gases under isothermal
Week 8	and adiabatic conditions for reversible and irreversible processes, enthalpy and internal energy change at constant P, V &T, Kirchhoff's equation.
24 th –29 th March	constant 1, 1 & 1, Entermon 5 equation.
30 th March	SUNDAY
Week 9	Second law of thermodynamics and its limitations, different statements of the law, Carnot's cycle and its efficiency, Carnot's theorem, thermodynamics scale of temperature. Concept of entropy—entropy as a state function, entropy change in ideal gases, , entropy as a criterion of spontaneity and equilibrium
31 st –5 th April	and equilibrium
6 th April	SUNDAY
o whin	DOMBIT

	Unit-III
Week 10	Hydrocarbons Alkanes: Physical and chemical properties of alkanes, free radical substitutions,
7 th –12 th April	
/ -12 April	halogenation, concept of relative reactivity v/s selectivity. Alkenes: Structure and isomerism, general
	methods of preparation, physical and chemical properties. Saytzeff and Hoffmann elimination,
	reactions: acidity, electrophilic and nucleophilic additions, hydration to form carbonyl compounds,
10th 4 17	alkylation of terminal alkynes.
13 th April	SUNDAY
	Mechanism of E1, E2, E1cb reactions, electrophilic addition (mechanism with suitable examples),
Week 11	Markownikoff rule, syn and anti-addition, addition of H2, X2 oxymercuration-demercuration,
a 4th a 0th a	hydroboration- oxidation, ozonolysis, hydroxylation. Alkynes: General methods of preparation of
14 th -19 th April	alkynes
20 th April	SUNDAY
	Unit-IV Aromatic Hydrocarbons and Dienes Concept of aromaticity, Huckel's rule, aromatic character
Week 12	of arenes, cyclic carbocations and carbanions with suitable examples and heterocyclic compounds with
21st-26th April	suitable examples, ,
6 th October	SUNDAY
	electrophilic aromatic substitution: halogenation, nitration, sulphonation, Friedel Crafts alkylation/
Week 13	acylation with their mechanism, directing effects of groups in electrophilic substitution
1st-8thFeb	
27 th April	SUNDAY
	nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of
Week 14	butadiene, chemical reactions- 1, 2 and 1, 4 additions (electrophilic and free radical mechanism), Diels –
28th-3rdMay	Alder reaction.
	SUNDAY
4 th May	
XX7 1 4 F	Revision Unit-1 and test
Week 15	
5th-10thMay	
510-Way	SUNDAY
11 th May	SUNDAI
11 May	Revision Unit-2 and test
Week 16	10 10 10 10 10 10 10 10 10 10 10 10 10 1
TO TO	
12th-17thMay	
1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1,	SUNDAY
18 th May	
	Revision Unit-3 and test
Week 17	
19th-24thMay	
,	SUNDAY
25 th May	
	Revision Unit-4 and test
Week 18	
26 th -31 st May	
26 th -31 st May	

Name of College: Pt. N.R.S. Government College, Rohtak

Academic Session:2024- 25 Semester: Even Name of Asstt./Ass.Prof: Praveen, Sandeep

Kumar

Class: B.Sc.6th sem (N.M, Med.) Sec A & B Name of Subject: Physical Chemistry

	Winter Break
Week 1	
1 st –6 th Jan	
	Unit 1: Introduction to Electronic Spectra, electronic transitions and band spectra, potential energy
Week 2	curves and Frank-Condon Principle
7 st –11 th Jan	
a a th	SUNDAY
12 th Jan	
Week 3	Electronic spectrum of a diatomic molecule, selection rules for electronic transitions in molecules, molecular orbitals involved in electronic transitions.
13st-18th Jan	
10th T	SUNDAY
19 th Jan	Unit-2: Introduction to photochemistry, interaction of radiation with matter, differences between
Week 4	thermal and photochemical processes, laws of photochemistry: Grothus-Draper Law, Stark-Einstein Law
20st-25th Jan	
a cth T	SUNDAY
26 th Jan	Jablonski Diagram depicting various processes occurring in the excited state, qualitative description
Week 5	of Fluorescence, Phosphorescence, non-radiative processes (internal conversion, inter-system crossing)
27 st –1 th Feb	
and E. I	SUNDAY
2 nd Feb	
	Quantum yield, Quenching of fluorescence, Photosensitization, Photo-inhibitors, Photochemical
Week 6	Equilibrium
3 rd -8 th Feb	
	SUNDAY
9 th Feb	
Week 7	Unit :- 3:Ideal and Non-ideal solutions, methods of expressing the concentration of solutions, activity and activity coefficient
10 st –15 th Feb	
16 th Feb	SUNDAY
Week 8	Dilute solutions, colligative properties, Raoult's law, relative lowering of vapour pressure, molecular weight determination
17 st –22 th Feb	
23 th Feb	SUNDAY

Osmosis law of osmotic proceurs, massurement of osmotic proceurs, determination of malacular	
Osmosis, law of osmotic pressure, measurement of osmotic pressure, determination of molecular weight from osmotic pressure	
SUNDAY	
Elevation in Boiling Point, Depression in Freezing Point, Thermodynamic derivation of	
relationship between molecular weight and elevation in boiling point and depression in freezing	
point	
Holi Break	
Experimental methods for determination of elevation in boiling point and depression in freezing	
point, abnormal molar masses, degree of dissociation and association of solutes.	
SUNDAY	
SUNDAI	
Unit-4. Statement and meaning of the terms- phase, component and degree of freedom	
Onte-4. Statement and meaning of the terms- phase, component and degree of freedom	
SUNDAY	
Thermodynamic derivation of Gibbs phase rule, phase equilibria of one- component system: Water system and Sulphur system.	
SUNDAY	
Phase equilibria of two component system, simple eutectic system: Pb-Ag system, desilverisation of	
lead	
SUNDAY	
Assignment 1,2,3	
And Test	
SUNDAY	
Revision	
IXCVISIOII	
SUNDAY	
Revision	

Name of College: Pt. N.R.S. Government College, Rohtak

Academic Session:2024-25 Semester: Even

Name of Asstt./Ass.Prof: Dr. Poonam,

Manoj

Class: B.Sc.4th sem (N.M, Med.) Sec A & B Name of Subject: Organic Chemistry

Week 1	Winter Break	
WEEK I		
1 st –6 th Jan		
Week 2	Molecular vibrations, Hooke's law, selection rules	
7 st –11 th Jan		
12th Ton	SUNDAY	
12 th Jan	intensity and position of ID hands massurament of ID spectrum. Fingarment	
Week 3	intensity and position of IR bands, measurement of IR spectrum, Fingerprint region	
13 st –18 th Jan		
19 th Jan	SUNDAY	
1) Jan	Applications of IR spectroscopy in structure elucidation of simple	
Week 4	Organic compounds	
20st-25thJan		
26th Ion	SUNDAY	
26 th Jan	Structure and nomanalature of aminos physical properties Caparation of a	
Week 5	Structure and nomenclature of amines, physical properties. Separation of a mixture of primary, secondary and tertiary amines.	
27 st –1 th Feb		
and E. I.	SUNDAY	
2 nd Feb	Structural features affecting basicity of amines. Prepa ration of alkyl and aryl	
Week 6	amines (reduction of nitro compounds, nitriles)	
3 rd -8 th Feb		
other 1	SUNDAY	
9 th Feb	Duananation of all rel and anyl aminos (node ative aminotion of ald sheet is and	
Week 7	Preparation of alkyl and aryl amines (reductive amination of aldehydic and ketonic compounds. Gabrielphthalimide reaction, Hofmann bromamide	
10st-15thFeb	reaction.	
16 th Feb	SUNDAY	
Week 8	Electrophilic aromatic substitution in aryl amines, reactions of amines with nitrous acid.	
17 st –22 th Feb		
23 th Feb	SUNDAY	
Week 9	Mechanism of diazotisation, structure of benzene diazonium chloride Replacement of diazo group by H, OH, F, Cl, Br, I, NO2 and CN groups	

24st-1stMarch	
	SUNDAY
2 nd March	
Week 10	Reduction of diazonium salts to hyrazines, coupling reaction and its synthetic application.
3 rd –8 th March	
Week 11 9 th -16 th March	Holi Break
Week 12	Preparation of nitro alkanes and nitro arenes and their chemical reactions.
17 th –22 th March	
23 th March	SUNDAY
Week 13	March Mechanism of electrophilic substitution reactions in nitro arenes and their
24 th –29 th March	reductions in acidic, neutral and alkaline medium.
30 th March	SUNDAY
Week14	Nomenclature and structure of the carbonyl group. Synthesis of aldehydes and ketones with particular reference to the synthesis of aldehydes from acid chlorides
31 st –5 th April 6 th April	SUNDAY
Week 15 7 th –12 th April 13 th April	Advantage of oxidation of alcohols with chromium trioxide (Sarett reagent) pyridinium chlorochromate (PCC) and pyridinium dichromate., Physical properties. Comparison of reactivities of aldehydes and ketones SUNDAY
Week 16	Mechanism of nucleophilic additions to carbonyl group with particular emphasis on benzoin, aldol, Perkin and Knoevenagel condensations. Condensation with ammonia and its derivatives
14 th -19 th April	
20 th April Week 17 21 st –26 th April	Wittig reaction. Mannich reaction. Oxidation of aldehydes, Baeyer–Villiger oxidation of ketones, Cannizzaro reaction. MPV, Clemmensen, Wolff-Kishner, LiAlH4 and NaBH4 reductions
27 th April	SUNDAY
Week 18 28 th –3 rd May	Assignment Test

Lesson Plan

B.Sc. Non Medical 4th Semester

1 taille of	Assistant Professor	Dr. Jyoti, Lokesh kumar
Class and Semester		B. Sc. Non Medical & non med
		Sec A,B(Semester – 4)
Subject		Chemistry
Paper		Inorganic Chemistry
Week -	lanthanide - Electroni	c structure, oxidation states
Week –	and ionic radii and lar	nthanide contraction,
Week -	complex formation, or	ecurrence and isolation, lanthanide compounds.
Week -	Actinides General features and c	hemistry of actinides,
Week - 5	chemistry of separation Am from U,	•
Week -	•	erties of Lanthanides and Actinides and with
6	transition elements.	
Week -	Revision	
Week -	Test of Actinides	
	•	and Quanti tative Inorganic Analysis-I of various acidic radicals,
	Chemistry of identific	
10	radicals in typical com	
Week -		nce of acid radicals including
11	•	alysis of basic radicals
Week -	Chemistry of interfere	nce of acid radicals including
12	•	alysis of basic radicals
Week –	Revision	
Week –	Theory of Qualitative	and Quantitative Inorganic Analysis-II
14	-	of various groups of basic radicals,
Week -	Theory of precipitation	
Week –	Post- precipitation, pu	rification of precipitate

16	
Week –	Revision
17	

Name of College: Pt. N.R.S. Government College, Rohtak

Academic Session:2024- 25 Semester: Even Name of Asstt./Ass.Prof: Dr.Poonam Devi

Class: B.Sc.Chem(H) 6th sem

Name of Subject: Inorganic Chemistry-II

	Winter Break
Week 1	
1 st –6 th Jan	
_	Air Pollution: Primary and secondary pollutants, sources, pollution effects and control
Week 2	
7 st –11 th Jan	
4 Ath T	SUNDAY
12 th Jan	pollution effects and control of the following: gaseous hydrocarbons, carbon monoxide
Week 3	pollution effects and control of the following, gaseous hydrocarbons, carbon molloxide
13st-18th Jan	SUNDAY
19 th Jan	SUNDAY
	carbon dioxide, hydrogen sulphide, oxides of sulfur and nitrogen and ozone, mechanism of
Week 4	photochemical smog formation; Air purification by micro organisms, Acid rain.
20 st –25 th Jan	
20 20 Jun	SUNDAY
26 th Jan	
Week 5	Water Pollution Types of water pollution, sources of water pollution, approaches to prevent and
VV CCR 5	control water pollution.
27 st –1 th Feb	
2 nd Feb	SUNDAY
Z Feb	Industrial Wastes and treatment processes: Introduction, characteristics of industrial wastes, types
Week 6	of industrial wastes,.
3 rd –8 th Feb	
5 6 Feb	SUNDAY
9 th Feb	
XX71- <i>7</i> 7	principles of industrial waste treatment and disposal of industrial wastes.
Week 7	
10st–15thFeb	
16 th Feb	SUNDAY
Week 8	Nuclear and Radio- Chemistry Composition of Nuclei, structure of nucleus, forces operative within nucleus, nuclear stability and mass energy equivalence (binding energy).
VI CCK O	nucleus, nuclear stability and mass energy equivalence (binding energy).
17 st –22 th Feb	
23 th Feb	SUNDAY
Week 9	Nuclear reactions: Types of nuclear reactions, the compound nucleus theories, thermonuclear reactions including fusion and fission reactions
TT CCR 7	reactions metading ration and resolutions

24 st –1 st March	
	SUNDAY
2 nd March	
	radiation detection and measurement: gaseous ion collection methods (G.M., ionisation and
Week 10	proportional counters) scientillation counter, semi - conductors detectors
3 rd -8 th March	
Week 11	Holi Break
9 th -16 th March	
W 1 12	Tracers in Chemistry Activation analysis, isotopic dilution analysis and radiometric titrations.
Week 12	
17 th –22 th March	
17 -22 March	SUNDAY
23 th March	
20 March	Crystal Structure: Structures of binary compounds such as zinc blende, wurtzite
Week 13	Crystal Structure. Structures of officiary compounds such as Zine Siende, wartzite
,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	
24 th –29 th March	
30 th March	SUNDAY
	Crystal Structure: Structures of binary compounds such as NiAs, CsCl, CaF2,
Week14	
31 st -5 th April	
6 th April	SUNDAY
	Crystal Structure: Structures of binary compounds such as rutile, β-Crystobalite, CdI2, BiI3
Week 15	
7 th –12 th April	
13 th April	SUNDAY
	Crystal Structure: Structures of binary compounds such as ReO3, corrundum and Mn2O3,
Week 16	
a 4th a 0th	
14 th -19 th April	CYNYDAY
20 th April	SUNDAY
	factors affecting crystal structures., Revision
Week 17	
21st-26th April	CYTNIDAY
27 th April	SUNDAY
*** • 40	Revision
Week 18	
28 th –3 rd May	

Summary of Lesson Plan of College Faculty
Name of College: Pt. N.R.S. Government College, Rohtak
Academic Session:2024- 25 Semester: Even

Name of Asstt./Ass.Prof: Dr.Poonam Devi Class: B.Sc.physics (H) 2nd sem (Minor) **Name of Subject: Minor Chemistry**

Name of Subj	ect: Minor Chemistry
	Unit–I
Week 1	Metal and Non-Metals
,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	Occurrence of elements in nature, physical and chemical properties of metals and non-metals, minerals
1st-8thFeb	and ores,
	SUNDAY
9 th Feb	
	metallurgical processes (benefaction, roasting, calcination and reduction of metal oxides processes),
Week 2	refining of metals, metallurgy of Fe, Zn, Al and Cu.
10st-15thFeb	
16 th Feb	SUNDAY
	Solution
Week 3	Types of solutions, expression of concentration of solutions of solids in liquids, solubility of gases in
17 st –22 th Feb	liquids,
23 th Feb	SUNDAY
25° Feb	solid solutions, Raoult's law, colligative properties - relative lowering of vapour pressure, elevation o
Week 4	boiling point,
VV CCK 4	boning point,
24 st –1 st March	
24 1 March	SUNDAY
2 nd March	
Week 5	depression in freezing point, osmotic pressure, determination of molecular masses using colligative
	properties, abnormal molecular mass, Van't Hoff factor
3 rd -8 th March	
Week 6	Holi Break
9 th -16 th March	
	Hydrocarbons
Week 7	Alkanes: General methods of preparation and Reactions: free radical substitution.
17 th –22 th March	
17 -22 March	SUNDAY
23 th March	SONDAI
25 March	Alkenes: General methods of preparation and Reactions: cis-addition (alk. KMnO4) and trans-addition
Week 8	(bromine),
VV CCK O	(oronnic),
24 th –29 th March	
30 th March	SUNDAY
	addition of HX (Markownikoff's and anti-Markownikoff's addition), hydration, ozonolysis,
Week 9	oxymecuration-demercuration, hydroboration oxidation.
31 st –5 th April	
6 th April	SUNDAY
	Alkynes: General methods of preparation and Reactions: formation of metal acetylides and acidity of
Week 10	alkynes, addition of bromine and alkaline KMnO4, ozonolysis and oxidation with hot alk. KMnO4,

-th anth /	
7 th –12 th April	hydration to form carbonyl compounds.
13 th April	SUNDAY
Week 11	Aromatic Hydrocarbons Structure of benzene (Kekule, hybrid and resonance), preparation of benzene.
14 th -19 th April	
20 th April	SUNDAY
Week 12 21st-26th April	preparation of benzene. Reactions: electrophilic substitution reactions in benzene citing examples of nitration,
6 th October	SUNDAY
Week 13 1st-8thFeb	halogenation, sulphonation and Friedel-Craft's alkylation and acylation
27 th April	SUNDAY
Week 14 28 th –3 rd May	acylation with special emphasis on carbocationic rearrangement, side chain oxidation of alkyl benzene.
4 th May	SUNDAY Revision Unit-1 and test
Week 15	Revision Unit-1 and test
5 th –10 th May	
11 th May	SUNDAY
Week 16	Revision Unit-2 and test
12 th -17 th May	
18 th May	SUNDAY
Week 17	Revision Unit-3 and test
19 th –24 th May	
25 th May	SUNDAY
Week 18	Revision Unit-4 and test
26th-31stMay	

Name of College: Pt. N.R.S. Government College, Rohtak

Academic Session:2024- 25 Semester: Even

Name of Asstt./Ass.Prof: Nidhi Mann &

Pinki Rani

Class: B.Sc.2nd sem

Name of Subject: Chemistry SEC Paper-II

Traine of Sun	ject: Chemistry SEC Paper-11
Week 1	Unit–I Solid Fuels Coal - origin, chemical composition, carbonization of coal, manufacture and properties of metallurgical coke, recovery of by-products
1 st –8 th Feb	
	SUNDAY
9 th Feb	
Week 2	calorific value, classification, characteristics & distribution of Indian coals
10st-15thFeb	
16 th Feb	SUNDAY
Week 3	storage and spontaneous combustion of coal, coal washing and blending, petrographic constituents of coal
17 st –22 th Feb	
23 th Feb	SUNDAY
Week 4	Unit–II Liquid Fuels Origin and composition of crude oil kerosene and diesel oil
24 st –1 st March	
2 nd March	SUNDAY
Week 5	crude oil distillation and its products with special reference to gasoline
3 rd –8 th March	
Week 6 9 th -16 th March	Holi Break
Week 7	cracking and reforming, coal tar distillation products, shale oil.
17 th –22 th March	
23 th March	SUNDAY
Week 8	Unit–III Gaseous Fuels
24 th –29 th March	Natural gas, coal gas, coke oven and blast furnace gas
30 th March	SUNDAY
Week 9	manufacture of water gas and producer gas
31 st –5 th April	
6 th April	SUNDAY

	carburetted water gas. Synthetic fuels: hydrogenation of coal, Fischer–Tropsch synthesis.
Week 10	
7 th –12 th April	
13 th April	SUNDAY
	Unit-IV Nuclear Fuels Introduction, nuclear fuels and nuclear reactors
Week 11	
14 th -19 th April	
20 th April	SUNDAY
	moderators and structural materials, introduction to renewable energy sources, Combustion: combustion
Week 12	of solids fuels
21st-26th April	
27 th April	SUNDAY
*	calculation of volume and weight of air necessary for combustion of fuels, gas analysis.
Week 13	
28th-3rdMay	
20 0 112	SUNDAY
4th May	
Tiviny	Revision Unit-1 and test .,
Week 14	Revision Unit-1 and test .,
WCCK 17	
5 th –10 th May	
5"-10 May	SUNDAY
1 1 th N / C	SUNDAY
11 th May	Desired Tirele 2 and 4ast
*** * 4 #	Revision Unit-2 and test
Week 15	
. ath a —tha g	
12 th –17 th May	
. a	SUNDAY
18 th May	
	Revision Unit-3 and test
Week 16	
19th-24thMay	
	SUNDAY
25 th May	
	Revision Unit-4 and test
Week 17	
26 th -31 st May	
<u> </u>	

Name of College: Pt. N.R.S. Government College, Rohtak

Academic Session:2024- 25 Semester: Even Name of Asstt./Ass.Prof:Abhishek Dahiya

Class: B.Sc.Chem(H) 4th sem

Name of Subject: Physical Chemistry

Traine of Sus	jett. Thysical Chemistry
	Winter Break
Week 1	Whiter Break
VV CCIX 1	
1st-6thJan	
	Thermodynamics – II Second law of thermodynamics. Need of the law, different definitions of the
Week 2	law, Carnot Cycle and its efficiency. Carnot theorem. Thermodynamic scale of temperature.
7 st –11 th Jan	
	SUNDAY
12 th Jan	
	Concept of entropy, entropy as a state function of V and T, entropy as a function of P and T. Entropy
Week 3	change in physical processes. Clausius inequility. Entropy as criteria of spontaneity and equilibrium
13st-18th Jan	
	SUNDAY
19 th Jan	
	Entropy change in ideal gases and mixing of gases, work function, Gibb,s free energy function.
Week 4	Gibbs function (G) and Helmholtz function (A) as thermodynamic function.
anst asthr	
20st-25th Jan	CLINID A N
26th Ion	SUNDAY
26 th Jan	
Week 5	Criteria of spontaneity of reversible processes in terms of enthalpy change, entropy change, work
Week 5	function and free energy function. Variation of G and A with P,V and T
27 st –1 th Feb	runction and free energy function. Variation of G and A with 1, V and 1
27 1 1 00	SUNDAY
2 nd Feb	SCIVERIT
100	Gibb Helmholtz equation and its application ,clausius- clapeyron equation Nernst heat theorem.
Week 6	Third law of thermodynamics and its applications. Partial molar quantities. Chemical potential.
	Gibb's Duhem equation. Gibb,s adsorption equation and its application, variation of chemical
3 rd -8 th Feb	potential with temperature and pressure.
	SUNDAY
9 th Feb	
	Redox reactions, electrolytic and galvanic cells. Reversible and irreversible cells reversible
Week 7	electrodes, types of reversible electrodes, metal electrodes, gas metal electrode, metal insoluble salt
	on ions and redox electrodes
10st-15thFeb	
16 th Feb	SUNDAY
	electrode reactions, cell voltage ,function of salt bridge, electrode potential and its determination.
Week 8	Standard hydrogen electrode, reference electrode, standard cell, sign convention. Electrochemical
	series and its significance.
17st–22thFeb	CVINIDAY
23 th Feb	SUNDAY
	Nernst equation for a reversible electrode and cell. Calculation of thermodynamic quantities of a cell
Week 9	reaction G, H and K
OAST ASTRE	
24 st –1 st March	

	SUNDAY
2 nd March	
	Polarisation over potential and hydrogen over voltage
Week 10	
3 rd -8 th March	
Week 11	Holi Break
9 th -16 th March	
TT 1 10	Experimental methods of chemical kinetics: conductometric, potentiometric, optical method
Week 12	polarimetery and spectrophotometer,
17 th –22 th March	
	SUNDAY
23 th March	
	Theories of reaction rates, effect of temperature on rate of reaction. Simple collision theory based
Week 13	upon transition state, hard sphere model theory (equilibrium hypothesis)
	apon unismon sime, muse spirote model intesty (equinosium hypomesis)
24 th –29 th March	
30 th March	SUNDAY
	Expression for the rate constants based on equilibrium constant their thermodynamic aspect
Week14	
31 st –5 th April	
6 th April	SUNDAY
	Definition of pH. Determination of pH using hydrogen, quinhydron and glass electrode by
Week 15	potentiometric method.
7 th –12 th April	
13 th April	SUNDAY
	Buffers solution, Buffer action, Henderson - Hazel equation.
Week 16	•
14 th -19 th April	
20 th April	SUNDAY
	Hydrolysis of salts, corrosion, types, theories and methods of controlling it.
Week 17	
21st-26th April	
27 th April	SUNDAY
	Revision.
Week 18	
28th-3rd May	

Name of College: Pt. N.R.S. Government College, Rohtak

Academic Session:2024- 25 Semester: Even Name of Asstt./Ass.Prof: Dr.Poonam Devi Class: B.Sc.physics (H) 2nd sem (Minor)

Name of Subject: Minor Chemistry

	ject: Minor Chemistry	
	Unit–I	
Week 1	Metal and Non-Metals	
_	Occurrence of elements in nature, physical and chemical properties of metals and non-metals, minerals	
1 st –8 th Feb	and ores,	
other I	SUNDAY	
9 th Feb		
Week 2	metallurgical processes (benefaction, roasting, calcination and reduction of metal oxides processes), refining of metals, metallurgy of Fe, Zn, Al and Cu.	
10st–15thFeb		
16 th Feb	SUNDAY	
	Solution	
Week 3	Types of solutions, expression of concentration of solutions of solids in liquids, solubility of gases in liquids,	
17 st –22 th Feb		
23 th Feb	SUNDAY	
	solid solutions, Raoult's law, colligative properties - relative lowering of vapour pressure, elevation of	
Week 4	boiling point,	
Ast Asta F		
24 st –1 st March	CLINID A X/	
2 nd March	SUNDAY	
Week 5	depression in freezing point, osmotic pressure, determination of molecular masses using colligative	
3 rd –8 th March	properties, abnormal molecular mass, Van't Hoff factor	
Week 6	Holi Break	
9 th -16 th March	Hon break	
7 10 Waren	Hydrocarbons	
Week 7	Alkanes: General methods of preparation and Reactions: free radical substitution.	
17 th –22 th March		
a a tha a	SUNDAY	
23 th March		
W I- O	Alkenes: General methods of preparation and Reactions: cis-addition (alk. KMnO4) and trans-addition	
Week 8	(bromine),	
24 th –29 th March		
30 th March	SUNDAY	
	addition of HX (Markownikoff's and anti-Markownikoff's addition), hydration, ozonolysis,	
Week 9	oxymecuration-demercuration, hydroboration oxidation.	
_		
31 st –5 th April		
6 th April	SUNDAY	
	Alkynes: General methods of preparation and Reactions: formation of metal acetylides and acidity of	
Week 10	alkynes, addition of bromine and alkaline KMnO4, ozonolysis and oxidation with hot alk. KMnO4,	
7 th –12 th April	hydration to form carbonyl compounds.	

13 th April	SUNDAY	
	Aromatic Hydrocarbons	
Week 11	Structure of benzene (Kekule, hybrid and resonance), preparation of benzene.	
14 th -19 th April		
20 th April	SUNDAY	
	preparation of benzene. Reactions: electrophilic substitution reactions in benzene citing examples of	
Week 12 21 st –26 th April	nitration,	
6 th October	SUNDAY	
	halogenation, sulphonation and Friedel-Craft's alkylation and acylation	
Week 13 1st–8thFeb		
27 th April	SUNDAY	
	acylation with special emphasis on carbocationic rearrangement, side chain oxidation of alkyl	
Week 14	benzene.	
28 th -3 rd May		
441-7	SUNDAY	
4 th May	D ' 11 ' 1 1 1 1	
Week 15	Revision Unit-1 and test	
5 th -10 th May		
	SUNDAY	
11 th May		
	Revision Unit-2 and test	
Week 16		
12 th –17 th May		
	SUNDAY	
18 th May		
Week 17	Revision Unit-3 and test	
19 th –24 th May		
•	SUNDAY	
25 th May		
Week 18	Revision Unit-4 and test	
26 th -31 st May		

Summary of Lesson Plan of College Faculty Name of College: Pt. N.R.S. Government College, Rohtak

Academic Session:2024- 25 Semester: Even Name of Asstt./Ass.Prof: Abhishek Dahiya

Class: B.Sc.Chem(H) 6th sem

Name of Subject: Physical Chemistry-II

	Winter Break
Week 1	
1 st –6 th Jan	
Week 2	
7 st –11 th Jan	
, 11 0011	SUNDAY
12 th Jan	
Week 3	
13 st –18 th Jan	
10th I	SUNDAY
19 th Jan	Construction of molecular orbital by linear combination of atomic orbital ,-H2 ion
Week 4	Construction of molecular orbital by linear combination of atomic orbital,-H2 ion
20st–25thJan	
- cu -	SUNDAY
26 th Jan	
Week 5	Molecular orbitals theory-II Calculation of energy levels from wave function, physical picture of bonding and antibonding wave function.
	picture of bonding and and conding wave function.
27 st –1 th Feb	
2 nd Feb	SUNDAY
Z FED	.Concept of pie, pie* orbitals and their characteristics. Hybrid orbital (Sp. Sp2 and Sp3 .).
Week 6	Calculation of co-efficients of atomic orbitals used in these hybrid orbitals.
3 rd -8 th Feb	
	SUNDAY
9 th Feb	
Week 7	Introduction of valence bond model of H2,comparison of molecular orbital. and valence bond. model
10 st –15 th Feb	
16 th Feb	SUNDAY
Week 8	Catalysis Homogeneous and Heterogeneous catalysis, Enzyme catalysis.
17 st –22 th Feb	

23 th Feb	SUNDAY	
20 100	Theory of catalysis - Intermediate compound formation theory, adsorption theory	
Week 9	Theory of educations intermediate compound formation theory, adsorption theory	
24 st –1 st March		
_	SUNDAY	
2 nd March		
Week 10	general characteristics of catalysis, positive catalysis, negative catalysis, autocatalysis, shape selective catalysis.	
3 rd –8 th March		
Week 11 9 th -16 th March	Holi Break	
Week 12	Chromatography Classification of chromatographic methods	
vv eek 12		
17 th –22 th March		
23 th March	SUNDAY	
Week 13	principle of differential migration, nature of differential migration	
24 th –29 th March		
30 th March	SUNDAY	
Week14	Adsorption phenomenon, nature of adsorpent, solvent system. Rf- values	
31 st –5 th April		
6 th April	SUNDAY	
Week 15 7 th –12 th April	application basic principle of partition, paper, column, thin layer liquid-liquid partition and high performance.	
13 th April	SUNDAY	
Week 16	Liquid chromatography, paper & column, thin layer liquid-liquid partition and high performance liquid chromatography	
14 th -19 th April		
20 th April	SUNDAY	
Week 17 21st-26th April	Revision	
27 th April	SUNDAY	
Week 18 28 th –3 rd May	Revision	

Name of College: Pt. N.R.S. Government College, Rohtak

Academic Session:2024- 25 Semester: Even Name of Asstt./Ass.Prof: Dr.Poonam Devi Class: B.Sc.physics (H) 2nd sem (Minor) Name of Subject: Minor Chemistry

Name of	Subject: Minor Chemistry	
*** 1 4	Unit-I	
Week 1	Metal and Non-Metals	
1 st –8 th Feb	Occurrence of elements in nature, physical and chemical properties of metals and non-metals, minerals	
1 -о гер	and ores, SUNDAY	
9 th Feb	BUNDAI	
7 1 0 0	metallurgical processes (benefaction, roasting, calcination and reduction of metal oxides processes),	
Week 2	refining of metals, metallurgy of Fe, Zn, Al and Cu.	
10 st –15 th Feb		
16 th Feb	SUNDAY	
	Solution	
Week 3	Types of solutions, expression of concentration of solutions of solids in liquids, solubility of gases in	
	liquids,	
17st-22thFeb		
23 th Feb	SUNDAY	
	solid solutions, Raoult's law, colligative properties - relative lowering of vapour pressure, elevation of	
Week 4	boiling point,	
Ast 1star 1		
24 st –1 st March	SUNDAY	
2 nd March	SUNDAT	
Week 5	depression in freezing point, osmotic pressure, determination of molecular masses using colligative	
VVCCRC	properties, abnormal molecular mass, Van't Hoff factor	
3 rd -8 th March	properties, abnormal molecular mass, vant from factor	
Week 6	Holi Break	
9 th -16 th March		
	Hydrocarbons	
Week 7	Alkanes: General methods of preparation and Reactions: free radical substitution.	
17 th –22 th March		
	SUNDAY	
23 th March		
	Alkenes: General methods of preparation and Reactions: cis-addition (alk. KMnO4) and trans-addition	
Week 8	(bromine),	
24 th –29 th March		
30 th March	SUNDAY	
SU March	addition of HX (Markownikoff's and anti-Markownikoff's addition), hydration, ozonolysis,	
Week 9	oxymecuration-demercuration, hydroboration oxidation.	
Week 5		
31 st –5 th April		
6 th April	SUNDAY	
	Alkynes: General methods of preparation and Reactions: formation of metal acetylides and acidity of	
Week 10	alkynes, addition of bromine and alkaline KMnO4, ozonolysis and oxidation with hot alk. KMnO4,	
7 th –12 th April	hydration to form carbonyl compounds.	

13 th April	SUNDAY
	Aromatic Hydrocarbons
Week 11	Structure of benzene (Kekule, hybrid and resonance), preparation of benzene.
14 th -19 th April	
20 th April	SUNDAY
	preparation of benzene. Reactions: electrophilic substitution reactions in benzene citing examples of
Week 12 21 st –26 th April	nitration,
6 th October	SUNDAY
	halogenation, sulphonation and Friedel-Craft's alkylation and acylation
Week 13 1st_8thFeb	
27 th April	SUNDAY
	acylation with special emphasis on carbocationic rearrangement, side chain oxidation of alkyl
Week 14	benzene.
28 th –3 rd May	
	SUNDAY
4 th May	
	Revision Unit-1 and test
Week 15	
5 th -10 th May	
	SUNDAY
11 th May	
v	Revision Unit-2 and test
Week 16	
12 th –17 th May	
_	SUNDAY
18 th May	
	Revision Unit-3 and test
Week 17	
19 th –24 th May	
	SUNDAY
25 th May	
	Revision Unit-4 and test
Week 18	
26 th -31 st May	

Lesson Plan

B.Sc. (Hons) VI Semester

Name of Assistant Professor		Dr. Rinki Malik
Class and Semester		B. Sc. (Hons).III Year VIth Semester
Subject		Chemistry
Paper		Paper- LVII (Theory) Organic Chemistry-II
Week 1	Introduction, esse	ential oils, classification of terpenes Isolation
1st-6thJan		
Week 2	Isoprene rule, iso	olation, structure elucidation
7 st -11 th Jan		
Week 3	Synthesis of citra	al and geraniol.
13 st –18 th Jan		
Week 4	Introduction, class alkaloids, general	ssification, extraction, physiological action in l characteristics
20st-25thJan		
Week 5		s of determining structures, Hofman's exhaustive ation, structure elucidation.
27st-1thFeb		
Week 6	Synthesis of nico	otine, cocaine, coniine and piperine.
3 rd -8 th Feb		
Week 7	Classification, Na Rotenoide, Saboo	atural pesticides: Nicotinides, Pyrethroids, dilia, Ryania.
10st-15thFeb	~	
Week 8	Synthetic pestic aromatic hydroca	cides: Nitrophenols, Halogens derivatives of arbons and alicyclic hydrocarbons,
17 st –22 th Feb		
Week 9	- 1	rus pesticides. Preparation, reactions Γ, BHC, Malathion and Parathion.
24st-1stMarch		
Week 10	Assignment -1,	Test & Discussion unit 1
3 rd -8 th March		

Week 11 9 th -16 th March	Holi Break	
Week 12	Introduction, classification, pro vitamins, occurrence.	
17 th -22 th March		
Week 13	Structure and deficiency diseases of vitamins A, B complex (B1, B2, B6 and B12), C, D, E, H and K	
24th-29th March		
Week14	Introduction, functions, difference between harmones and vitamins,	
31st-5th April		
Week 15 7 th –12 th April	classification and study of Thyroxine, Adrenalin, Insulin,	
Week 16	Testosterone, Progesterone, Estrogens, Discussion & Revision Unit 2	
14 th -19 th April		
Week 17 21 st –26 th April	Cortison (structure, secreting gland and functions) Discussion & Revision Unit 3	
Week 18 28 th –3 rd May	Discussion & Revision Unit 4	

Lesson Plan

B.Sc. 2nd Semester

Name of Assi	stant Professor	Dr. Rinki Malik
Class and Semester		B.Sc. (Chemistry as Single Major)
Subject		Chemistry, SEC Paper – II
		Nomenclature :Fuel Chemistry
Paper		Course Code 24CHE402SE0
Week 1		n, chemical composition, calorific value, aracteristics & distribution of Indian coals,
1st-8thFeb		
Week 2		ntaneous combustion of coal, coal washing and raphic constituents of coal.
10st-15thFeb		
Week 3		of coal, manufacture and properties of te, recovery of by-products
17st-22thFeb	0	
Week 4	Origin and comp	osition of crude oil, crude oil distillation.
24st-1stMarch		
Week 5	Products with spoil,	pecial reference to gasoline,kerosene and diesel
3 rd -8 th March		
Week 6 9 th -16 th March	Holi Break	
Week 7	Cracking and refe	orming, coal tar distillation products, shale oil.
17 th –22 th March		
Week 8	Natural gas, coal	gas, coke oven and blast furnace gas.
24 th –29 th March		
Week 9	Manufacture of v	vater gas and producer gas, carburetted water gas.
31st_5th April		
Week 10 7 th –12 th April	Synthetic fuels: h	nydrogenation of coal, Fischer–Tropsch synthesis.

Week 11	Assignment-1 Test -1, Discussion & Revision.
14 th -19 th April	
Week 12 21 st –26 th April	Introduction, nuclear fuels and nuclear reactors, moderators and structural materials.
Week 13 28 th –3 rd May	Introduction to renewable energy sources.
Week 14	Combustion: combustion of solids fuels,
5 th –10 th May	
Week 15	Calculation of volume and weight of air necessary for combustion of fuels, Assignment-2 Test -2, Discussion & Revision
12 th -17 th May	
Week 16	Gas analysis study, Discussion & Revision Unit 2 and 3
19th-24th May	
Week 17	Discussion & Revision Unit 4
26 th -31 st May	

Name of College: Pt. N.R.S. Government College, Rohtak

Academic Session:2024- 25 Semester: Even

Name of Asstt./Ass.Prof: Reena, Neeraj Class: B.Sc.6th sem (N.M, Med.) Sec A & B Name of Subject: Organic Chemistry

	Winter Break	
Week 1	willer Break	
1 st –6 th Jan		
Week 2	Introduction: Molecular orbital picture and aromatic characteristics of pyrrole, furan, thiophene and pyridine.	
7 st –11 th Jan		
anth T	SUNDAY	
12 th Jan		
Week 3	Methods of synthesis and chemical reactions with particular emphasis on the mechanism of electrophilic substitution.	
13 st –18 th Jan		
10th T	SUNDAY	
19 th Jan	M. 1	
Week 4	Mechanism of nucleophilic Substitution reactions in pyridine derivatives.	
20st-25thJan		
	SUNDAY	
26 th Jan		
Week 5	Comparison of basicity of pyridine, piperidine and pyrrole	
27 st –1 th Feb		
	SUNDAY	
2 nd Feb		
Week 6	Introduction to condensed five and six-membered heterocycles.	
3 rd -8 th Feb		
	SUNDAY	
9 th Feb		
Week 7	Prepration and reactions of indole, quinoline and isoquinoline with special reference to Fisher indole synthesis	
10 st –15 th Feb		
16 th Feb	SUNDAY	
Week 8	Skraup synthesis and Bischler-Napieralski synthesis.	
17 st –22 th Feb		
23 th Feb	SUNDAY	
Week 9	Mechanism of electrophilic substitution reactions of quinoline and isoquinoline	

24 st –1 st March	
	SUNDAY
2 nd March	
Week 10	Nomenclature, structural features, Methods of formation and Chemical
	reactions of thiols, thioethers, sulphonic acids, sulphonamides and
3 rd –8 th March	sulphaguanidine.
Week 11	Holi Break
9 th -16 th March	
Week 12	Synthetic detergents alkyl and aryl sulphonates.
17 th –22 th March	
	SUNDAY
23 th March	
Week 13	Acidity of α -hydrogens, alkylation of diethyl malonate and ethyl acetoacetate.
24th–29thMarch	
30 th March	SUNDAY
	Synthesis of ethyl acetoacetate: the Claisen condensation. Keto-enol
Week14	tautomerism of ethyl acetoacetate. Addition or chain-growth polymerization.
31 st –5 th April	Free radical vinyl polymerization, ionic vinyl polymerization, Ziegler-Natta polymerization and vinyl polymers
6 th April	SUNDAY
Week 15 7 th –12 th April	Condensat ion or step growth polymerization. Polyeste rs ,polyamides, phenol formaldehyde resins, urea formaldehyde resins,epoxy resins and polyurethanes. Natural and synthetic rubbers
13 th April	SUNDAY
Week 16	Classification, of amino acids. Acid-base behavior, isoelectric point and electrophoresis. Preparation of amino acids.
14 th -19 th April	
20 th April	SUNDAY
Week 17	Structure and nomenclature of peptides and proteins. Classification of proteins.
21st–26th April	Peptide structure determination, end group analysis, selective hydrolysis of
	peptides.
27 th April	SUNDAY
r	Assignment
Week 18 28 th –3 rd May	Test

Summary of Lesson Plan of College Faculty Name of College: Pt. N.R.S. Government College, Rohtak

Semester: Even Academic Session:2024- 25 Name of Asstt./Ass.Prof: Nidhi Mann, Anil

& Rinku

Class: B.Sc.2nd sem (DSC-II) Sec A, B &

 \mathbf{C}

Name of Subject: Chemistry

Name of Sub	ject: Chemistry
	Unit-I
Week 1	Non-aqueous Solvents Physical properties of a solvent, types of solvents and their general
1 st –8 th Feb	characteristics, solvent system concept
1 -6 Feb	SUNDAY
9 th Feb	SCI-DIT
Week 2	reactions in non-aqueous solvents with reference to liquid NH3 and liquid SO2. Hard and soft acids and bases (HSAB concept), applications of HSAB principle
10 st –15 th Feb	
16 th Feb	SUNDAY
	nature of bonding in noble gas compounds (valence bond treatment and MO treatment for XeF2 and
Week 3	XeF4), molecular shapes of noble gas compounds (VSEPR theory). Noble Gases Occurrence and uses
17 st –22 th Feb	
23 th Feb	SUNDAY
Week 4	, rationalization of inertness of noble gases, clathrates, preparation and properties, chemical properties of the noble gases, chemistry of xenon: structure and bonding in xenon fluorides, oxides and oxyfluorides (XeF2, XeF4, XeF6, XeO3, XeO4, XeOF2, XeO2F2, XeOF4, XeF5 + , XeF5 -)
24 st –1 st March	
	SUNDAY
2 nd March	
Week 5	Unit-II Thermodynamics Brief discussion upto first law of thermodynamics, heat capacity, heat capacities at constant volume and pressure and their relationship.
3 rd –8 th March	constant volume and pressure and then retailousing.
Week 6	Holi Break
9 th -16 th March	
Week 7	Joule's law, Joule—Thomson coefficient for ideal gases and real gases and inversion temperature entropy as a function of V & T, entropy as a function of P & V.
17 th –22 th March	
	SUNDAY
23 th March	
Week 8	calculation of work and heat, dU & dH for the expansion of ideal gases and real gases under isothermal and adiabatic conditions for reversible and irreversible processes, enthalpy and internal energy change at
24 th –29 th March	constant P, V &T, Kirchhoff's equation.
30 th March	SUNDAY
Week 9	Second law of thermodynamics and its limitations, different statements of the law, Carnot's cycle and its efficiency, Carnot's theorem, thermodynamics scale of temperature. Concept of entropy—entropy as a state function, entropy change in ideal gases, , entropy as a criterion of spontaneity and equilibrium
31 st –5 th April	and open and
6 th April	SUNDAY
l E	

	Unit-III
Week 10	Hydrocarbons Alkanes: Physical and chemical properties of alkanes, free radical substitutions,
7 th –12 th April	
/ -12 April	halogenation, concept of relative reactivity v/s selectivity. Alkenes: Structure and isomerism, general
	methods of preparation, physical and chemical properties. Saytzeff and Hoffmann elimination,
	reactions: acidity, electrophilic and nucleophilic additions, hydration to form carbonyl compounds,
10th 4	alkylation of terminal alkynes.
13 th April	SUNDAY
	Mechanism of E1, E2, E1cb reactions, electrophilic addition (mechanism with suitable examples),
Week 11	Markownikoff rule, syn and anti-addition, addition of H2, X2 oxymercuration-demercuration,
a 4th a 0th a	hydroboration- oxidation, ozonolysis, hydroxylation. Alkynes: General methods of preparation of
14 th -19 th April	alkynes
20 th April	SUNDAY
	Unit-IV Aromatic Hydrocarbons and Dienes Concept of aromaticity, Huckel's rule, aromatic character
Week 12	of arenes, cyclic carbocations and carbanions with suitable examples and heterocyclic compounds with
21st–26th April	suitable examples, ,
6 th October	SUNDAY
	electrophilic aromatic substitution: halogenation, nitration, sulphonation, Friedel Crafts alkylation/
Week 13	acylation with their mechanism, directing effects of groups in electrophilic substitution
1st-8thFeb	
27 th April	SUNDAY
	nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of
Week 14	butadiene, chemical reactions- 1, 2 and 1, 4 additions (electrophilic and free radical mechanism), Diels –
28th-3rdMay	Alder reaction.
	SUNDAY
4 th May	
XX7 1 4 5	Revision Unit-1 and test
Week 15	
5 th -10 th May	
510-Way	SUNDAY
11 th May	SUNDAI
11 May	Revision Unit-2 and test
Week 16	10 10 10 10 10 10 10 10 10 10 10 10 10 1
TT CON IU	
12th-17thMay	
	SUNDAY
18 th May	
	Revision Unit-3 and test
Week 17	
19th-24thMay	
,	SUNDAY
25 th May	
•	Revision Unit-4 and test
Week 18	
26 th -31 st May	

Name of College: Pt. N.R.S. Government College, Rohtak

Academic Session:2024-25 Semester: Even Name of Asstt./Ass.Prof: Bhupender Singh,

Rekha Gautam

Class: B.Sc.6th sem (N.M, Med.) Sec A & B Name of Subject: Inorganic Chemistry

	Winter Break
Week 1	
1 st –6 th Jan	
_	Organometallic Chemistry Definition, nomenclature and classification of organometallic compounds
Week 2	Preparation.
7 st –11 th Jan	
12th Ton	SUNDAY
12 th Jan	Properties, and bonding of alkyls of Li, Al, Hg, and Sn a brief account of metal-ethylenic complexes.
Week 3	1 species, and conding of any is of 21, 12, 12g, and on a cite account of mount cary terms compressed.
13 st –18 th Jan	
13 –16 Jan	SUNDAY
19 th Jan	
Week 4	Mononuclear carbonyls.
20st-25th Jan	CLINID ANZ
26 th Jan	SUNDAY
	The nature of bonding in metal carbonyls.
Week 5	
27 st –1 th Feb	
and E. I	SUNDAY
2 nd Feb	
Week 6	Acids and Bases, HSAB Concept Arrhenius, Bronsted – Lowry, the Lux – Flood, Solvent system.
Week 0	
3 rd -8 th Feb	
9 th Feb	SUNDAY
7 100	Lewis concepts of acids & Dases, relative strength of acids & Dases.
Week 7	
10st–15thFeb	
16 th Feb	SUNDAY
Week 8	Concept of Hard and Soft Acids & Dases.
VV CCK O	
17st–22thFeb	CUINIDAN
23 th Feb	SUNDAY

	Disinguis Chamistan Francis and the state in his large all manages
Week 9	Bioinorganic Chemistry, Essential and trace elements in biological processes.
WEEK 9	
24 st –1 st March	
	SUNDAY
2 nd March	
	Metalloporphyrins with special reference to haemoglobin and myoglobin.
Week 10	
and others	
3 rd -8 th March	Holi Break
Week 11 9 th -16 th March	Holl Break
9 -10 Maich	Biological role of alkali and alkaline earth metal ions with special reference to Ca2+.
Week 12	Biological fole of alkali and alkaline cartif metal folis with special reference to Ca2+.
17 th –22 th March	
	SUNDAY
23 th March	
	Nitrogen fixation.
Week 13	
24th 20thMonch	
24 th –29 th March 30 th March	SUNDAY
50 March	Sil icones and Phosphazenes Silicones and phosphazenes, their preparation.
Week14	on reones and r nospitazones officines and phospitazones, their preparation.
,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	
31 st –5 th April	
6 th April	SUNDAY
	Structure and uses of Silicones and phosphazenes.
Week 15	
7 th –12 th April	
13 th April	SUNDAY
Week 16	Assignment 1,2,3
Week 16	And Test
14 th -19 th April	
20 th April	SUNDAY
12h111	Revision
Week 17	
21st-26th April	
27 th April	SUNDAY
	Revision
Week 18	
28 th -3 rd May	