

**LESSON PLAN- B.Sc(Hons.)1<sup>st</sup> SEMESTER****Session: 2024-25**

Name of teacher- Dr. Dayawati

Subject- DSC PAPER I General Chemistry-I

| WEEKS                  | SYLLABUS   |
|------------------------|--|
| 22-7-2024 to 27-7-2024 | Atomic Structure and Periodicity of Elements: Bohr's atomic model and its application,   |
| 29-7-2024 to 3-8-2024  | quantum numbers, their application and rules of electronic configuration, effective nuclear charge,  |
| 5-8-2024 to 10-8-2024  | shielding or screening effect, Slater rules, variation of effective nuclear charge in periodic table.  |
| 12-8-2024 to 17-8-2024 | Periodic trends in atomic radii, ionic radii and its calculation, covalent radii, electronegativity, electron gain enthalpy, ionization enthalpy and factors affecting ionization energy. Pauling, Mullikan and Allred Rachow scales.  |
| 20-8-2024 to 24-8-2024 | Ionic Solids: Ionic bond and its characteristics and factors affecting, types of Bravais lattice, voids, packing in solids, determination of radius ratio of all voids, radius ratio rule and its limitations.<br>Packing of ions in crystals, calculation of density and crystal structures of ionic solids (NaCl, CsCl, ZnS, CaF <sub>2</sub> , Na <sub>2</sub> O), defect structures in crystal. Born-Landé equation with derivation, expression for lattice energy, Madelung constant, Born-Haber cycle and its application with examples, solvation energy. |
| 27-8-2024 to 31-8-2024 | Semiconductors, types of semiconductors, valence bond and band theories (alloys excluded).<br>Gaseous State-I: Elementary treatment of gas laws, kinetic gas equation and its derivation,<br>deviations from ideal gas behaviour, compressibility factor (Z) and its variation with pressure and temperature for different gases,  |

|                          |   |
|--------------------------|---|
| 2-9-2024 to 7-9-2024     | Van der Waals equation of state, its derivation and application in explaining real gas behavior, mention other equations of state (Berthelot, Dielectric or Dieterici), Van der Waals equation expressed in virial form and calculation of Boyle temperature, critical temperature, critical pressure, critical volume and their determination. |
| 9-9-2024 to 14-9-2024    | Isotherms of real gases and their comparison with Van der Waals isotherms, continuity of states, relationship between critical constants and Van der Waals constants, law of corresponding states, reduced equation of state.   |
| 16-9-2024 to 21-9-2024   | Basics of Organic Chemistry and Stereochemistry: Electronic displacements and their applications, reactive intermediates, types of organic reactions and energy considerations.   |
| 23-9-2024 to 28-9-2024   | Methods of determination of reaction mechanism (product analysis, intermediates, isotope effects, kinetic and stereochemical studies).  |
| 30-9-2024 to 5-10-2024   | Stereoisomerism: Optical activity and optical isomerism, asymmetry, chirality, enantiomers, diastereomers.  |
| 7-10-2024 to 12-10-2024  | Specific rotation, configuration and projection formulae: Newmann, Sawhorse, Fischer and their interconversion.   |
| 14-10-2024 to 19-10-2024 | Chirality in molecules with one and two stereocentres: meso configuration, racemic mixture and their resolution.  |
| 21-10-2024 to 26-10-2024 | Relative and absolute configuration: D/L and R/S designations.  |
| 4-11-2024 to 9-11-2024   | designations. Geometrical isomerism: cis-trans, syn-anti and E/Z notations using CIP rules.   |
| 11-11-2024 to 20-11-2024 | Assignments, Viva, Test, Revision   |
| 23-11-2024 to 20-12-2024 | MDU examination   |
| 21-12-2024 to 31-12-2024 | Winter break  |

**LESSON PLAN- B.Sc(Hons.)1<sup>st</sup> SEMESTER****Session: 2024-25**

Name of teacher- Dr. Ravinder Singh

Subject- DSC PAPER II General Chemistry-II

| WEEKS                  | SYLLABUS  |
|------------------------|---|
| 22-7-2024 to 27-7-2024 | Chemical Bonding: Covalent bonds and their types, valence shell electron pair repulsion theory (VSEPR etc.). Valence bond theory (Heitler-London approach), and its limitations.  |
| 29-7-2024 to 3-8-2024  | Bent rule, covalent character in ionic compounds, polarizing power and polarizability. Dipole moment and its applications, Fajan's rules and its applications. Molecular orbital theory and MO diagrams of heteronuclear diatomic molecules (CO, NO, HCl, CO <sub>2</sub> , HF etc) (idea of s-p mixing and orbital interaction to be given). |
| 5-8-2024 to 10-8-2024  | Gaseous State II: Degree of freedom and principle of equipartition of energy, Maxwell's distribution law of molecular velocities and energies, root mean square velocity, average velocity and most probable velocity and their relationship.   |
| 12-8-2024 to 17-8-2024 | Mean free path and its derivation, collision diameter, collision number and collision frequency, viscosity of gases and effect of temperature and pressure on viscosity of gases, relationship between mean free path and coefficient of viscosity,   |
| 20-8-2024 to 24-8-2024 | calculation of molecular diameter from coefficient of viscosity. Liquid State: Structure of liquids, properties of liquids – surface tension, refractive index, viscosity, vapour pressure and optical rotation   |

|                          |   |
|--------------------------|---|
| 27-8-2024 to 31-8-2024   | Alkanes and Alkenes Carbon-Carbon Sigma Bond: Chemistry of alkanes, methods of preparation of alkanes, physical and chemical properties. Free radical substitution: Halogenation – relative reactivity and selectivity. |
| 2-9-2024 to 7-9-2024     | Carbon-Carbon Pi Bond: Structure and isomerism, general methods of preparation, mechanism of E1, E2, E1cB reactions, Saytzeff and Hoffmann eliminations. Reactions of alkenes: Electrophilic additions,                 |
| 9-9-2024 to 14-9-2024    | Markownikoff rule, syn and anti-addition, addition of H <sub>2</sub> X <sub>2</sub> , oxymercuration-demercuration, hydroboration-oxidation, ozonolysis, hydroxylation and polymerization.                              |
| 16-9-2024 to 21-9-2024   | Dienes: Classification, methods of preparation, chemical reaction: Diels Alder reaction, 1,2- and 1,4-addition reactions in conjugated dienes.  |
| 23-9-2024 to 28-9-2024   | Mechanism of allylic and benzylic bromination in propene, 1-butene, toluene, ethyl benzene.   |
| 30-9-2024 to 5-10-2024   | Alkynes: General methods of preparation, reactions of alkynes: acidity, electrophilic and nucleophilic additions, hydration to form carbonyls. Aromatic Hydrocarbons:   |
| 7-10-2024 to 12-10-2024  | Concept of aromaticity, Huckel rule, aromatic character of arenes,  |
| 14-10-2024 to 19-10-2024 |   |
| 21-10-2024 to 26-10-2024 | cyclic carbocations and carbanions with suitable examples and heterocyclic compounds with suitable examples.  |
| 4-11-2024 to 9-11-2024   | Electrophilic aromatic substitution: halogenation, nitration, sulphonation, Friedel Crafts alkylation/acylation with their mechanism. Directing effects of groups in electrophilic substitution reactions.              |

|                          |                                   |
|--------------------------|-----------------------------------|
| 11-11-2024 to 20-11-2024 | Assignments, Viva, Test, Revision |
| 23-11-2024 to 20-12-2024 | MDU examination                   |
| 21-12-2024 to 31-12-2024 | Winter break                      |

| <b>LESSON PLAN- B.Sc1<sup>st</sup> SEMESTER          Session: 2024-25</b> |  |
|---|--|
| Name of teacher- Dr. Suman Bhatti, Dr. Pinki, Dr. Poonam Devi             |  |
| Class: B.Sc.1st sem(MINOR)  |  |
| Name of Subject: Basic Concepts of Chemistry Paper 1                      |  |
| <b>22<sup>nd</sup>-27<sup>th</sup> July</b>                               | <b>Atomic Structure:</b> Atomic Models, Rutherford's model and its limitations, Bohr's model and its applications.                   |
| <b>28<sup>th</sup> July</b>   | <b>SUNDAY</b>  |
| <b>29<sup>th</sup>-03<sup>rd</sup> August</b>                             | Dual nature of matter and light, De Broglie's relationship, Heisenberg uncertainty principle.  |
| <b>04<sup>th</sup> August</b>   | <b>SUNDAY</b>  |
| <b>5<sup>th</sup>-10<sup>th</sup> August</b>                              | Concept of orbitals, quantum numbers, shapes of s, p and d orbitals,   |
| <b>11<sup>th</sup> August</b>   | <b>SUNDAY</b>  |
| <b>12<sup>th</sup>-17<sup>th</sup> August</b>                             | Rules for filling electrons in orbitals - Aufbau principle, Pauli's exclusion principle and Hund's rule                              |
| <b>18<sup>th</sup> August</b>   | <b>SUNDAY</b>  |
| <b>19<sup>th</sup>-24<sup>th</sup> August</b>                             | Electronic configuration of atoms, stability of half-filled and completely filled orbitals. Shapes of s, p, d orbitals.              |
| <b>25<sup>th</sup> August</b>   | <b>SUNDAY</b>  |
| <b>26<sup>th</sup>-31<sup>st</sup> August</b>                             | <b>Periodic table and atomic properties:</b> Brief history of the development of periodic table, modern periodic law and the present |

|   |   |
|---|---|
|   | form of periodic table,.  |
| <b>1<sup>st</sup> September</b>                     | <b>SUNDAY</b>   |
| <b>2<sup>nd</sup> -7<sup>th</sup> September</b>     | Periodic trends in properties of elements -atomic radii, ionic radii, inert gas radii.  |
| <b>08<sup>th</sup> September</b>                    | <b>SUNDAY</b>   |
| <b>9<sup>th</sup>-14<sup>th</sup> September</b>     | Periodic trends in properties of elements -Ionization enthalpy, electron gain enthalpy, electronegativity, valency. Nomenclature of elements with atomic number greater than 100. |
| <b>15<sup>th</sup> September</b>                    | <b>SUNDAY</b>   |
| <b>16<sup>th</sup>-21<sup>st</sup> September</b>    | <b>Mole concept:</b> Atomic and molecular masses, mole concept and molar mass.  |
| <b>22<sup>nd</sup> September</b>                    | <b>SUNDAY</b>   |
| <b>23<sup>th</sup>-28<sup>th</sup> September</b>    | Avogadro's number and its significance, percentage composition, empirical and molecular formula.  |
| <b>29<sup>th</sup> September</b>                    | <b>SUNDAY</b>   |
| <b>30<sup>th</sup> Sep -05<sup>th</sup> October</b> | Chemical reactions, Solution preparations (Molarity, Normality, molality, mole percentage, strength)Stoichiometric calculations involving reactants and product.                  |
| <b>06<sup>th</sup> October</b>                      | <b>SUNDAY</b>   |
| <b>07<sup>th</sup>-12<sup>th</sup> October</b>      | Fundamentals of Organic Chemistry: Electronic displacements: Inductive effect, electromeric effect, resonance, hyperconjugation.  |
| <b>13<sup>th</sup> October</b>                      | <b>SUNDAY</b>   |
| <b>14<sup>th</sup>-19<sup>th</sup> October</b>      | Cleavage of bonds: homolysis and heterolysis. Reaction intermediates: carbocations, carbanions.   |

|   |  |
|---|--|
| <b>20<sup>th</sup> October</b>                      | <b>SUNDAY</b>  |
| <b>21<sup>st</sup>-26<sup>th</sup> October</b>      | Reaction intermediates: free radicals, and carbenes. Electrophiles and nucleophiles. |
| <b>27<sup>th</sup> October</b>                      | <b>SUNDAY</b>  |
| <b>4<sup>th</sup> -9<sup>th</sup> November</b>      | Aromaticity: benzenoids and Huckel's rule.   |
| <b>10<sup>th</sup> November</b>                     | <b>SUNDAY</b>  |
| <b>11<sup>th</sup> -16<sup>th</sup> November</b>    | Revision and test  |
| <b>17<sup>th</sup> November</b>                     | <b>SUNDAY</b>  |
| <b>18<sup>th</sup> November onwards till Exams.</b> | Test discussion  |

|   |   |
|---|---|
| <b>LESSON PLAN- B.Sc1<sup>st</sup> SEMESTER      Session: 2024-25</b> |   |
| Name of teacher- Dr. Rinki , Dr. Pinki , Nidhi Mann                   |   |
| Class: B.Sc.1st Sem(SEC)  |   |
| Subject-Role of Chemistry in Society                                  |   |
| <b>22<sup>nd</sup>-27<sup>th</sup> July</b>                           | <b>Analysis of soil: Composition of soil</b>  |
| <b>28<sup>th</sup> July</b>   | <b>SUNDAY</b>   |
| <b>29<sup>th</sup>-03<sup>rd</sup> August</b>                         | Concept of pH and pH measurement of soil  |
| <b>04<sup>th</sup> August</b>   | <b>SUNDAY</b>   |
| <b>5<sup>th</sup>-10<sup>th</sup> August</b>                          | Complexometric titrations, Chelation, Chelating agents, use of indicators   |
| <b>11<sup>th</sup> August</b>   | <b>SUNDAY</b>   |
| <b>12<sup>th</sup>-17<sup>th</sup> August</b>                         | Estimation of Calcium and Magnesium ions in soil.   |
| <b>18<sup>th</sup> August</b>   | <b>SUNDAY</b>   |
| <b>19<sup>th</sup>-24<sup>th</sup> August</b>                         | Analysis of water: Definition of pure water, sources responsible for contaminating water, water sampling methods. |
| <b>25<sup>th</sup> August</b>   | <b>SUNDAY</b>   |
| <b>26<sup>th</sup>-31<sup>st</sup> August</b>                         | water purification methods. Determination dissolved oxygen of a water sample.                                     |
| <b>1<sup>st</sup> September</b>                                       | <b>SUNDAY</b>   |

|   |  |
|---|--|
| <b>2<sup>nd</sup> -7<sup>th</sup> September</b>     | A general study including preparation and uses of the Hair dye, soap, shampoo.   |
| <b>08<sup>th</sup> September</b>                    | <b>SUNDAY</b>  |
| <b>9<sup>th</sup>-14<sup>th</sup> September</b>     | Preparation and uses of the suntan lotions, face powder, lipsticks, talcum powder, nail enamel.  |
| <b>15<sup>th</sup> September</b>                    | <b>SUNDAY</b>  |
| <b>16<sup>th</sup>-21<sup>st</sup> September</b>    | General introduction to pesticides (natural and synthetic), benefits and adverse effects   |
| <b>22<sup>nd</sup> September</b>                    | <b>SUNDAY</b>  |
| <b>23<sup>th</sup>-28<sup>th</sup> September</b>    | Changing concepts of pesticides, brief introduction of structure activity relationship   |
| <b>29<sup>th</sup> September</b>                    | <b>SUNDAY</b>  |
| <b>30<sup>th</sup> Sep -05<sup>th</sup> October</b> | Synthesis and technical manufacture and uses of representative pesticides in the Organochlorines (Gammexene,); Organophosphates (Malathion). |
| <b>06<sup>th</sup> October</b>                      | <b>SUNDAY</b>  |
| <b>07<sup>th</sup>-12<sup>th</sup> October</b>      | Basic principle of pH metric, potentiometric and conductometric titrations   |
| <b>13<sup>th</sup> October</b>                      | <b>SUNDAY</b>  |
| <b>14<sup>th</sup>-19<sup>th</sup> October</b>      | Applications of conductivity measurements: determination of degree of dissociation   |

|   |   |
|---|---|
| <b>20<sup>th</sup> October</b>                      | <b>SUNDAY</b>   |
| <b>21<sup>st</sup>-26<sup>th</sup> October</b>      | Determination of $K_a$ of acids and base, Buffer solution, Buffer action, |
| <b>27<sup>th</sup> October</b>                      | <b>SUNDAY</b>   |
| <b>4<sup>th</sup> -9<sup>th</sup> November</b>      | Henderson – Hazel equation, Buffer mechanism of buffer action.            |
| <b>10<sup>th</sup> November</b>                     | <b>SUNDAY</b>   |
| <b>11<sup>th</sup> -16<sup>th</sup> November</b>    | Revision and test   |
| <b>17<sup>th</sup> November</b>                     | <b>SUNDAY</b>   |
| <b>18<sup>th</sup> November onwards till Exams.</b> | Test discussion   |

**LESSON PLAN- B.ScChem. Hons. 3<sup>rd</sup> SEMESTER**

**Session: 2024-25**

Name of teacher- Abhishek

Class- B.Sc. Chemistry Hons. 3<sup>rd</sup> SEM

Subject- Physical Chemistry

| WEEKS                  | SYLLABUS  |
|------------------------|---|
| 22-7-2024 to 27-7-2024 | <b>Chemical Equilibrium</b><br>Types of Reactions (Reversible and irreversible) Equilibrium state. Le-chatelier principle.  |
| 29-7-2024 to 3-8-2024  | Law of mass action and its application to derive the law of chemical equilibrium. Thermodynamically derivation of law of chemical equilibrium.  |
| 5-8-2024 to 10-8-2024  | Equilibrium constant and free energy function, isotherms and reaction isochor, Classius -Claperon equation and its application.   |
| 12-8-2024 to 17-8-2024 | <b>Revision and Assignment</b>  |
| 20-8-2024 t 24-8-2024  | <b>Distribution Law</b><br>Nernst distribution law, Thermodynamic derivation of Nernst distribution law. Conditions for the validity of Nernst distribution law.  |
| 27-8-2024 to 31-8-2024 | Derivation of molecular complexity from distribution law. Application of distribution law i.e. calculation of solubility of solute in solvent.  |
| 2-9-2024 to 7-9-2024   | Determination of extent of association and dissociation of solute in the solvent, distribution indicator, process of extraction and determination of degree of hydrolysis and study of complex ion formation. |

|                          |   |
|--------------------------|---|
| 9-9-2024 to 14-9-2024    | <b>Thermodynamics-I</b><br>Important terms used in thermodynamic system, surrounding, type of system intensive and extensive property, state and path function and their differentials, thermodynamic equilibrium thermodynamic process |
| 16-9-2024 to 21-9-2024   | Concept of heat and work, first law of thermodynamics, (statement and derivation). Internal energy and enthalpy, internal energy and enthalpy change and their relation.  |
| 23-9-2024 to 28-9-2024   | Heat capacity. Heat capacity at constant volume and pressure and their relationship. Joule-Thomson effect and inversion temperature.  |
| 30-9-2024 to 5-10-2024   | Calculation of W, Q, $\Delta v$ and the expansion of ideal gas under isothermal and adiabatic conditions for reversible processes.  |
| 7-10-2024 to 12-10-2024  | <b>Revision, Test, Assignment</b>   |
| 14-10-2024 to 19-10-2024 | <b>Colloidal States:</b><br>Colloids, classification of colloids, solids in liquids (sols) properties: Kinetic, optical and electrical; stability of colloids   |
| 21-10-2024 to 26-10-2024 | Protective colloids Hardy-Schulze Rule, gold number, Emulsion types of emulsion and their preparation, Emulsifier.  |
| 4-11-2024 to 9-11-2024   | <b>Gels (liquid in solids):</b><br>Classification and properties, inhibition and general application of colloids  |
| 11-11-2024 to 20-11-2024 | <b>Revision, Test, Assignment</b>   |
| 23-11-2024 to 20-12-2024 | <b>MDU examination</b>  |
| 21-12-2024 to 31-12-2024 | <b>Winter break</b>   |

**LESSON PLAN- B.Sc Chem. Hons. 3<sup>rd</sup> SEMESTER**

**Session: 2024-25**

Name of teacher- Dr. Ravinder Singh  
Class- B.Sc. Chemistry Hons. 3<sup>rd</sup> SEM  
Subject- Organic Chemistry

| WEEKS                  | SYLLABUS   |
|------------------------|--|
| 22-7-2024 to 27-7-2024 | <b>Ultraviolet (UV) absorption spectroscopy</b><br>Ultraviolet (UV) absorption spectroscopy, absorption laws (Beer-Lambert law), molar absorptivity, presentation and analysis of UV spectra, types of electronic transitions, effect of conjugation. Concept of chromophore and auxochrome. |

|                          |  |
|--------------------------|--|
|                          | Bathochromic, hypsochromic, hyperchromic and hypochromic shifts.   |
| 29-7-2024 to 3-8-2024    | Woodward -Fieser rules, calculation of $m$ and $x$ of simple conjugated dienes   |
| 5-8-2024 to 10-8-2024    | Woodward -Fieser rules, calculation of $m$ and $x$ of unsaturated ketones. UV spectra of conjugated enes, enones, dienones, unsaturated acids, unsaturated esters, lactones, unsaturated amides and lactams.   |
| 12-8-2024 to 17-8-2024   | <b>Revision, Test, Assignment</b>  |
| 20-8-2024 to 24-8-2024   | <b>Alcohols</b><br>Classification and nomenclature. Monohydric alcohols, nomenclature, methods of formation by reduction of aldehydes, ketones, carboxylic acids and esters. Hydrogen bonding. Acidic nature.  |
| 27-8-2024 to 31-8-2024   | <b>Alcohols</b><br>Classification and nomenclature. Monohydric alcohols, nomenclature, methods of formation by reduction of aldehydes, ketones, carboxylic acids and esters. Hydrogen bonding. Acidic nature.  |
| 2-9-2024 to 7-9-2024     | Dihydric alcohols — nomenclature, methods of formation, chemical reactions of vicinal glycols, oxidative cleavage [ $\text{Pb}(\text{OAc})_4$ and $\text{HIO}_4$ ] and pinacol-pinacolone rearrangement  |
| 9-9-2024 to 14-9-2024    | <b>Phenols</b><br>Nomenclature, structure and bonding. Preparation of phenols, physical properties and acidic character. Comparative acidic strengths of alcohols and phenols, resonance stabilization of phenoxide ion.   |
| 16-9-2024 to 21-9-2024   | <b>Phenols</b><br>Reactions of phenols — electrophilic aromatic substitution, acylation and carboxylation. Mechanisms of Fries rearrangement, Claisen rearrangement, Gatterman synthesis, Hauben-Hoesch reaction, Lederer -Manasse reaction and Reimer - Tiemann reaction. |
| 23-9-2024 to 28-9-2024   | <b>Ethers and Epoxides</b><br>Nomenclature of ethers and methods of their formation, physical properties. Chemical reactions — cleavage and autoxidation, Ziesel's method. Synthesis of epoxides.  |
| 30-9-2024 to 5-10-2024   | <b>Ethers and Epoxides</b><br>Acid and base -catalyzed ring opening of epoxides, orientation of epoxide ring opening, reactions of Grignard and organolithium reagents with epoxides   |
| 7-10-2024 to 12-10-2024  | <b>Revision, Test, Assignment</b>  |
| 14-10-2024 to 19-10-2024 | <b>Carboxylic Acids &amp; Derivatives</b><br>Nomenclature, structure and bonding, physical properties, acidity of carboxylic acids, effects of substituents on acid strength.  |
| 21-10-2024 to 26-10-2024 | Methods of formation and chemical reactions of halo acids. Hydroxy acids: malic, tartaric and citric acids.  |

|                          |   |
|--------------------------|---|
| 4-11-2024 to 9-11-2024   | Structure and nomenclature of acid chlorides, esters, amides (urea) and acid anhydrides. Relative stability of acyl derivatives. Physical properties, interconversion of acid derivatives by nucleophilic acyl substitution.<br>Preparation of carboxylic acid derivatives, chemical reactions.<br>Mechanisms of esterification and hydrolysis (acidic and basic) |
| 11-11-2024 to 20-11-2024 | <b>Revision, Test, Assignment</b>   |
| 23-11-2024 to 20-12-2024 | <b>MDU examination</b>  |
| 21-12-2024 to 31-12-2024 | <b>Winter break</b>   |

**LESSON PLAN- B.Sc(Hons.)3rd SEMESTER**

**Session: 2024-25**

Name of teacher- **Dr. Anju Siwach**, Assistant Professor

Subject- Inorganic Chemistry I

| WEEKS                  | SYLLABUS   |
|------------------------|--|
| 22-7-2024 to 27-7-2024 | <b>Co-ordination Compounds:</b> General introduction   |
| 29-7-2024 to 3-8-2024  |  |
| 5-8-2024 to 10-8-2024  | Werner's coordination theory and its experimental verification,<br>effective atomic number concept, chelates,                        |
| 12-8-2024 to 17-8-2024 | nomenclature of coordination compounds isomerism in coordination compounds,<br><br>valence bond theory of transition metal complexes |

|                          |   |
|--------------------------|---|
| 20-8-2024 to 24-8-2024   | <p><b>Oxidation and Reduction:</b> Use of redox potential data - analysis of redox cycle,</p> <p>redox stability in water - Frost, Latimer and Pourbaix diagrams, Principles involved in the extraction of elements.</p> <p><b>Non-aqueous solvents</b> Physical properties of solvent, types of solvents and their general characteristics,</p> <p>reactions in non-aqueous solvents with reference to liquid NH<sub>3</sub> and liquid SO<sub>2</sub>.</p>  |
| 27-8-2024 to 31-8-2024   |   |
| 2-9-2024 to 7-9-2024     |   |
| 9-9-2024 to 14-9-2024    |   |
| 16-9-2024 to 21-9-2024   | <p><b>Chemistry of Elements of First Transition Series-I</b> Definition, characteristic properties of d-block elements. Properties of the elements of the first transition series,</p> <p>their binary compounds and complexes illustrating relative stability of their oxidation states, coordination number and geometry.</p>   |
| 23-9-2024 to 28-9-2024   |   |
| 30-9-2024 to 5-10-2024   | <p><b>Chemistry of Elements of First Transition Series-II</b> Chemistry of Ti, V, Cr, Mn, Fe and Co in various oxidation states.</p> <p>Titanium – oxides, oxyions, peroxides and halides</p> <p>Vanadium – halides, oxides, vanadates and vanadyl compounds</p> <p>Chromium – halides, oxides, chromates &amp; oxyhalides</p> <p>Manganese – oxides, permanganates, halides &amp; acetates</p> <p>Iron – oxides and iron compounds</p> <p>Cobalt – oxides, sulphates, halides and Co(III) complexes.</p> |
| 7-10-2024 to 12-10-2024  |   |
| 14-10-2024 to 19-10-2024 |   |
| 21-10-2024 to 26-10-2024 |   |
| 4-11-2024 to 9-11-2024   |   |
| 11-11-2024 to 20-11-2024 | <p>Assignments, Viva, Test, Revision</p> <p>MDU examination</p> <p>Winter break</p>   |
| 23-11-2024 to 20-12-2024 |   |
| 21-12-2024 to 31-12-2024 |   |

**LESSON PLAN- B.Sc(Hons.) 5<sup>th</sup> SEMESTER      Session: 2024-25**Name of teacher- **Dr. Ravinder Singh**

Subject- Organic Chemistry I

|   |  |
|---|--|
| <b>22<sup>nd</sup>-27<sup>th</sup> July</b>   | <b>Spectroscopy</b><br>Principle of nuclear magnetic resonance, the PMR spectrum, number of signals, peak areas, equivalent and nonequivalent protons  |
| <b>28<sup>th</sup> July</b>                   | <b>SUNDAY</b>  |
| <b>29<sup>th</sup>-03<sup>rd</sup> August</b> | Positions of signals and chemical shift, shielding and deshielding of protons, proton counting, splitting of signals and coupling constants, magnetic equivalence of protons.<br>Revision, Assignment, Test  |
| <b>04<sup>th</sup> August</b>                 | <b>SUNDAY</b>  |
| <b>5<sup>th</sup>-10<sup>th</sup> August</b>  | Discussion of PMR spectra of the molecules: ethyl bromide, n-propyl bromide, isopropyl bromide, 1,1-dibromoethane, 1,1,2-tribromoethane  |
| <b>11<sup>th</sup> August</b>                 | <b>SUNDAY</b>  |
| <b>12<sup>th</sup>-17<sup>th</sup> August</b> | Discussion of PMR spectra of the molecules: ethanol, acetaldehyde, ethyl acetate, toluene, benzaldehyde, acetophenone, <i>p</i> -anisidine and <i>p</i> -nitrotoluene. Simple problems on PMR spectroscopy for structure determination of organic compounds.<br>Revision, Assignment, Test |
| <b>18<sup>th</sup> August</b>                 | <b>SUNDAY</b>  |
| <b>19<sup>th</sup>-24<sup>th</sup> August</b> | <b>Mass Spectroscopy:</b> Introduction, instrumentation, mass spectrum, determination of molecular formula, parent peak and base peak, recognition of molecular ion peak.  |
| <b>25<sup>th</sup> August</b>                 | <b>SUNDAY</b>  |
| <b>26<sup>th</sup>-31<sup>st</sup> August</b> | <b>Mass Spectroscopy:</b> Fragmentation pattern of alkanes, alkenes and benzene.   |
| <b>1<sup>st</sup> September</b>               | <b>SUNDAY</b>  |

|   |   |
|---|---|
| <b>2<sup>nd</sup> -7<sup>th</sup> September</b>     | <b>OrganosulphurCompounds</b><br>Nomenclature, structural features, Methods of formation and chemical reactions of thiols, thioethers, sulphonic acids.   |
| <b>08<sup>th</sup> September</b>                    | <b>SUNDAY</b>   |
| <b>9<sup>th</sup>-14<sup>th</sup> September</b>     | <b>OrganosulphurCompounds</b><br>Methods of formation and chemical reactions of sulphonamides and sulphaguanidine. Synthetic detergents alkyl and aryl sulphonates.<br>Revision, Assignment, Test |
| <b>15<sup>th</sup> September</b>                    | <b>SUNDAY</b>   |
| <b>16<sup>th</sup>-21<sup>st</sup> September</b>    | <b>Carbohydrates</b><br>Classification and nomenclature. Monosaccharides, mechanism of formation, interconversion of glucose and fructose.  |
| <b>22<sup>nd</sup> September</b>                    | <b>SUNDAY</b>   |
| <b>23<sup>th</sup>-28<sup>th</sup> September</b>    | <b>Carbohydrates</b><br>Chain lengthening and chain shortening of aldoses. Configuration of monosaccharides. Erythro and threo diastereomers.   |
| <b>29<sup>th</sup> September</b>                    | <b>SUNDAY</b>   |
| <b>30<sup>th</sup> Sep -05<sup>th</sup> October</b> | <b>Carbohydrates</b><br>Open chain and cyclic structure of D(+)-glucose & D(-)fructose. Mechanism of mutarotation.  |
| <b>06<sup>th</sup> October</b>                      | <b>SUNDAY</b>   |
| <b>07<sup>th</sup>-12<sup>th</sup> October</b>      | <b>Carbohydrates</b><br>Structures of ribose and deoxyribose.   |
| <b>13<sup>th</sup> October</b>                      | <b>SUNDAY</b>   |
| <b>14<sup>th</sup>-19<sup>th</sup> October</b>      | <b>Carbohydrates</b>  |

|   |  |
|---|--|
|   | An introduction to disaccharides (maltose, sucrose and lactose)  |
| <b>20<sup>th</sup> October</b>                      | <b>SUNDAY</b>  |
| <b>21<sup>st</sup>–26<sup>th</sup> October</b>      | <b>Carbohydrates</b><br>An introduction to polysaccharides (starch and cellulose) without involving structure determination.   |
| <b>27<sup>th</sup> October</b>                      | <b>SUNDAY</b>  |
| <b>4<sup>th</sup> –9<sup>th</sup> November</b>      | <b>Organometallic Compounds</b><br>Organomagnesium compounds: the Grignard reagents - formation, structure and chemical reactions.<br>Organolithium compounds: formation and chemical reactions.<br>Organozinc compounds: formation and chemical reactions.<br>Organolead compounds: formation and chemical reactions. |
| <b>10<sup>th</sup> November</b>                     | <b>SUNDAY</b>  |
| <b>11<sup>th</sup> -16<sup>th</sup> November</b>    | <b>Organometallic Compounds</b><br>Organocadmium compounds: formation and chemical reactions.<br>Organocopper compounds: formation and chemical reactions<br>Revision, Assignment, Test  |
| <b>17<sup>th</sup> November</b>                     | <b>SUNDAY</b>  |
| <b>18<sup>th</sup> November onwards till Exams.</b> | Revision, Assignment, Test   |

|   |  |
|---|--|
| <b>LESSON PLAN- B.Sc(Hons.) 5<sup>th</sup> SEMESTER      Session: 2024-25</b> |  |
| Name of teacher- <b>Dr. Anju Siwach</b>                                       |  |
| Subject- Organic Chemistry II   |  |
| <b>22<sup>nd</sup>–27<sup>th</sup> July</b>                                   | Heterocyclic<br>Introduction: Molecular orbital picture and aromatic characteristics of pyrrole, furan, thiophene and pyridine.  |
| <b>28<sup>th</sup> July</b>   | <b>SUNDAY</b>  |
| <b>29<sup>th</sup>-03<sup>rd</sup> August</b>                                 | Methods of synthesis and chemical reactions with particular emphasis on the mechanism of electrophilic substitution. Mechanism of nucleophilic substitution reactions in pyridine derivatives. |
| <b>04<sup>th</sup> August</b>   | <b>SUNDAY</b>  |
| <b>5<sup>th</sup>-10<sup>th</sup> August</b>                                  | Comparison of basicity of pyridine, piperidine and pyrrole.  |

|   |   |
|---|---|
| <b>11<sup>th</sup> August</b>                       | <b>SUNDAY</b>   |
| <b>12<sup>th</sup>-17<sup>th</sup> August</b>       | Introduction to condensed five and six- membered heterocycles. Preparation and reactions of indole, quinoline and isoquinoline with special reference to Fisher indole synthesis  |
| <b>18<sup>th</sup> August</b>                       | <b>SUNDAY</b>   |
| <b>19<sup>th</sup>-24<sup>th</sup> August</b>       | Skraup synthesis and Bischler-Napieralski synthesis. Mechanism of electrophilic substitution reactions of indole, quinoline and isoquinoline  |
| <b>25<sup>th</sup> August</b>                       | <b>SUNDAY</b>   |
| <b>26<sup>th</sup>-31<sup>st</sup> August</b>       | Organo Phosphorus Compounds Introduction.   |
| <b>1<sup>st</sup> September</b>                     | <b>SUNDAY</b>   |
| <b>2<sup>nd</sup> -7<sup>th</sup> September</b>     | Nomenclature, Trivalent phosphorus compounds - trialkyl and triaryl phosphine (method of formation and reactions), Pentavalent phosphorus compounds, organic phosphoranes phosphorus ylides wittig reaction. Biological role of phosphorus.   |
| <b>08<sup>th</sup> September</b>                    | <b>SUNDAY</b>   |
| <b>9<sup>th</sup>-14<sup>th</sup> September</b>     | Polymers: Brief history of macromolecular Science Natural polymers: Starch, cellulose silk resin Classification, types of polymerization Addition, condensation and their mechanisms (free radical, ionic and coordination - Ziegler Natta Catalyst), methods of polymerisation - bulk suspension, emulsion and solution. |
| <b>15<sup>th</sup> September</b>                    | <b>SUNDAY</b>   |
| <b>16<sup>th</sup>-21<sup>st</sup> September</b>    | Detailed study of following polymers with respect to synthesis, properties and applications.<br>(I) Phenol formaldehydes resins. (II) Urea formaldehydes resins. (III) Polyesters (IV) Polyamides. (V) Natural and synthetic rubbers.   |
| <b>22<sup>nd</sup> September</b>                    | <b>SUNDAY</b>   |
| <b>23<sup>th</sup>-28<sup>th</sup> September</b>    | 1. Organic Synthesis via Enolates -hydrogens, alkylation of diethyl malonate and ethyl $\alpha$ -Acidity of acetoacetate. Synthesis of ethyl acetoacetate: the Claisen condensation. Keto-enol tautomerism of ethyl acetoacetate  |
| <b>29<sup>th</sup> September</b>                    | <b>SUNDAY</b>   |
| <b>30<sup>th</sup> Sep -05<sup>th</sup> October</b> | Alkylation of 1,3-dithianes. Alkylation and acylation of enamines Synthetic Dyes - Colour and constitution (electronic concept). Classification of dyes. Chemistry and synthesis of Methyl orange, Congo red, Malachite green, Crystal violet, Phenolphthalein, Fluorescein Alizarin and Indigo                           |
| <b>06<sup>th</sup> October</b>                      | <b>SUNDAY</b>   |
| <b>07<sup>th</sup>-12<sup>th</sup> October</b>      | Amino Acids, Peptides, Proteins and Nucleic Acids Classification, structure and stereochemistry of amino acids. Acid- base behavior, isoelectric point and electrophoresis.   |
| <b>13<sup>th</sup> October</b>                      | <b>SUNDAY</b>   |
| <b>14<sup>th</sup>-19<sup>th</sup> October</b>      | Preparation -amino acids. $\alpha$ and reactions of alpha -amino acids.   |

Name of teacher- Dr. Anju Siwach  
 Class- B.Sc. Chemistry Hons. 5<sup>th</sup> SEM  
 Subject- Inorganic Chemistry I

| WEEKS                                | SYLLABUS  |
|--------------------------------------|---|
| 22-7-2024<br>to 27-7-<br>2024        | <b>Unit 1: Metal - ligand Bonding in Transition Metal Complexes:</b> Valence bond theory and imitation of valence bond theory                                   |
| 29-7-2024<br>to 3-8-<br>2024         | An elementary idea of crystal-field theory, crystal field splitting in octahedral   |
| 5-8-2024<br><br>to 10-8-<br><br>2024 | Crystal field splitting in tetrahedral and square planar complexes  |
| 12-8-2024<br>to 17-8-<br>2024        | Factors affecting the crystal-field parameters. <b>Revision</b>   |
| 20-8-2024 t<br>24-8-2024             | <b>Unit 2: Magnetic Properties of Transition Metal complexes:</b> Types of magnetic behaviour, methods of determining magnetic susceptibility                   |
| 27-8-2024 to<br>31-8-2024            | Spin-only formula and L-S coupling, correlation of $\chi_s$ and $\chi_{eff}$ values   |
| 2-9-2024 to<br>7-9-2024              | orbital contribution to magnetic moments, application of magnetic moment data for 3d-metal complexes  |
| 9-9-2024 to<br>14-9-2024             | <b>Thermodynamic and Kinetic Aspects of Metal Complexes:</b> A brief outlines of thermodynamic stability of metal complexes                                     |
| 16-9-2024 to<br>21-9-2024            | Factors affecting the stability, substitution reactions of square planar complexes. <b>Test and Revision</b>  |
| 23-9-2024 to<br>28-9-2024            | <b>Unit 3: Electron Spectra of Transition Metal Complexes:</b> Types of electronic transitions, selection rules of d-d transitions, spectroscopic ground states |
| 30-9-2024 to<br>5-10-2024            | Spectrochemical series, Orgel - energy level diagram for d1 and d9 states, discussion of the electronic spectrum of $[Ti(H_2O)_6]^{3+}$                         |

|                          |  |
|--------------------------|--|
|                          | complex ion.   |
| 7-10-2024 to 12-10-2024  | <b>Hard and Soft Acids and Base (HSAB)</b> Classification of acids and bases as hard and soft. Pearson's HSAB concept, acid-base strength and hardness and softness. |
| 14-10-2024 to 19-10-2024 | Symbiosis, theoretical basis of hardness and softness, electronegativity and hardness and softness. <b>Assignment</b>  |
| 21-10-2024 to 26-10-2024 | <b>Unit 4: Silicons, Phosphazenes and S - N compounds:</b> Synthesis, properties nature of bonding, structures and applications of Silicons                          |
| 4-11-2024 to 9-11-2024   | Synthesis, properties nature of bonding, structures and applications of Phosphazenes   |
| 11-11-2024 to 20-11-2024 | Synthesis, properties nature of bonding, structures and applications of S - N compounds <b>Test and Revision</b>   |
| 23-11-2024 to 20-12-2024 | <b>MDU examination</b>   |
| 21-12-2024 to 31-12-2024 | <b>Winter break</b>  |

| <b>LESSON PLAN- B.Sc 5<sup>TH</sup> SEMESTER Session: 2024-25</b>   |   |
|---|---|
| Name of teacher- Dr. Rinki<br>Class- B.Sc. Chemistry Hons. 5 <sup>th</sup> SEM<br>Subject- Inorganic Chemistry II |   |
| WEEKS   | SYLLABUS  |
| 22-7-2024 to 27-7-2024  | <b>Unit 1: Organometallic Chemistry-I</b> :Introduction of Organometallic Chemistry, Definition, Nature of Metal Carbon bond and some characteristics                               |
| 29-7-2024 to 3-8-2024   | classification of organometallic compounds by bond types I )covalent ii) Ionic iii) Electron deficient iv) cluster compounds v) $\pi$ bond compounds including sandwich derivatives |
| 5-8-2024 to 10-8-2024   | Bonding in, metalethylenic, metal-acetylenic complexes,   |
| 12-8-2024 to 17-8-2024  | <b>Revision and Assignment</b>  |
| 20-8-2024 to 24-8-2024  | <b>Unit 2: Organometallic Chemistry-II:</b> Structure and   |

|                          |  |
|--------------------------|--|
|                          | bonding in Metal carbonyls, cyclopentadienyl derivative  |
| 27-8-2024 to 31-8-2024   | Applications of organometallic compounds as homogeneous catalysts in hydrogenation, hydroformylation   |
| 2-9-2024 to 7-9-2024     | Applications of organometallic compounds in polymerization, oligomerization  |
| 9-9-2024 to 14-9-2024    | Applications of organometallic compounds in alkynes and Ziegler - Natta polymerization of ethylene and propylene   |
| 16-9-2024 to 21-9-2024   | Metathesis reactions of alkenes <b>Test and Revision</b>   |
| 23-9-2024 to 28-9-2024   | <b>Unit 3: Bio- Inorganic Chemistry:</b> Essential and Trace elements in biological processes,   |
| 30-9-2024 to 5-10-2024   | Bioinorganic chemistry of haemoglobin and myoglobin,   |
| 7-10-2024 to 12-10-2024  | vitamin B <sub>12</sub> , carboxypeptidase A and chlorophyll,  |
| 14-10-2024 to 19-10-2024 | Biological role of alkali and alkaline earth metal ions with nitrogen fixation (special reference to Ca <sup>2+</sup> )<br><b>Assignment and Revision</b>    |
| 21-10-2024 to 26-10-2024 | <b>Unit 4: Medicinal Chemistry:</b><br>Medicinal aspects of some metal complexes - platinum metal complexes as anticanceragents and their probable mechanism |
| 4-11-2024 to 9-11-2024   | Anticancer activity of cu, Co and Au complexes. Antibacterial and antiviral activity of metal complexes.   |
| 11-11-2024 to 20-11-2024 | <b>Corrosion and Passivity:</b> Theories of corrosion, prevention of corrosion of metals, passivity <b>Test and Revision</b>                                 |
| 23-11-2024 to 20-12-2024 | <b>MDU examination</b>   |
| 21-12-2024 to 31-12-2024 | <b>Winter break</b>  |

**LESSON PLAN- B.Sc 5<sup>th</sup> SEMESTER**
**Session: 2024-25**

Name of teacher- Dr.Suman Bhatti  
Class- B.Sc. Chemistry Hons. 5<sup>th</sup> SEM  
Subject- Physical Chemistry I

| WEEKS                    | SYLLABUS   |
|--------------------------|--|
| 22-7-2024 to 27-7-2024   | Solution and collective - properties Ideal and Non-ideal solution. Methods of expressing concentrations of solution, activity and activity coefficient. Dilute solution. Colligative properties, Raoult's law. Relative lowering of vapour pressure. |
| 29-7-2024 to 3-8-2024    | Dilute solution. Colligative properties, Raoult's law. Relative lowering of vapour pressure.   |
| 5-8-2024 to 10-8-2024    | Molecular weight determination, osmotic law of osmotic pressure and its measurements. Determination of molecular weight by osmotic pressure method.  |
| 12-8-2024 to 17-8-2024   | Elevation of boiling point and depression in freezing point. Thermodynamic derivation of relation between molecular weight and elevation in boiling point and depression in freezing point.  |
| 20-8-2024 to 24-8-2024   | Experimental methods for determining various colligative properties. Abnormal molar mass. Degree of dissociation and association of solutes.   |
| 27-8-2024 to 31-8-2024   | Rotational Spectroscopy Introduction of electromagnetic radiations, regions of the spectrum, basic features of different spectrometers.  |
| 2-9-2024 to 7-9-2024     | Statement of the Born-Oppenheimer approximation, degree of freedom of diatomic molecule. Energy level of a rigid rotor (semiclassical principle) selection rule, spectral intensity.   |
| 9-9-2024 to 14-9-2024    | Distribution using population distribution (Maxwell-Boltzmann distribution) determination of bond length, qualitative description of nonrigid rotator. Isotopic effect   |
| 16-9-2024 to 21-9-2024   | Phase equilibrium Statement and meaning of the terms phase, component and degree of freedom. Phase rule and its thermodynamic derivation,  |
| 23-9-2024 to 28-9-2024   | Phase equilibria of one component system, water and sulfur system, phase equilibria of two component system, solid-liquid equilibria, simple eutetic (Bi-Cd; Pb-silver system), Desilverisation of lead.   |
| 30-9-2024 to 5-10-2024   | Solid solution: Compound formation with congruent melting point (Mg-Cu) and incongruent melting point (NaCl-Cu) (FeCl <sub>3</sub> and CuSO <sub>4</sub> - H <sub>2</sub> O) system freezing mixture, acetone, dry ice                               |
| 7-10-2024 to 12-10-2024  | Photo Chemistry: Interaction of radiation with matter. Photochemical reactions and their difference with thermal reaction law of photo chemistry.  |
| 14-10-2024 to 19-10-2024 | Grothuss, Drapper law Stark Einstein law, Lambert law, Beer's law  |
| 21-10-2024 to 26-10-2024 | Jablonski diagram depicting various processes occurring in the excited state qualitative description of Fluorescence,  |

|                          |  |
|--------------------------|--|
|                          |  |
| 4-11-2024 to 9-11-2024   | phosphorescence non-radiation processes (internal conversion, inter system crossing) quantum yield photosensitized reactions energy transfer processes (some simple examples). |
| 11-11-2024 to 20-11-2024 | Unit Test  |
| 23-11-2024 to 20-12-2024 | <b>MDU examination</b>   |
| 21-12-2024 to 31-12-2024 | <b>Winter break</b>  |

| <b>LESSON PLAN- B.Sc 5<sup>th</sup> SEMESTER      Session: 2024-25</b> |  |
|--|--|
| Name of teacher- Dr. Abhishek  |  |
| <b>Class:</b> B.Sc. 5 <sup>th</sup> Semester (Chem. Hons.)             |  |
| <b>Name of Subject:</b> Physical Chemistry-II                          |  |
| <b>22<sup>nd</sup>-27<sup>th</sup> July</b>                            | <b>Statistical Thermodynamics:</b><br>Statistical thermodynamics of Maxwell Boltzmann distribution law.                                  |
| <b>28<sup>th</sup> July</b>  | <b>SUNDAY</b>  |
| <b>29<sup>th</sup>-03<sup>rd</sup> August</b>                          | <b>Statistical Thermodynamics:</b><br>Maxwell-Boltzmann law and the concept of negative temperature.                                     |
| <b>04<sup>th</sup> August</b>  | <b>SUNDAY</b>  |
| <b>5<sup>th</sup>-10<sup>th</sup> August</b>                           | <b>Statistical Thermodynamics:</b><br>Maxwell-Boltzmann law of distribution of energy and velocity (evaluation of energy).               |
| <b>11<sup>th</sup> August</b>  | <b>SUNDAY</b>  |
| <b>12<sup>th</sup>-17<sup>th</sup> August</b>                          | <b>Statistical Thermodynamics:</b><br>Derivation of equation of states for a monatomic ideal gas.  |
| <b>18<sup>th</sup> August</b>  | <b>SUNDAY</b>  |
| <b>19<sup>th</sup>-24<sup>th</sup> August</b>                          | <b>Nuclear Chemistry and Radioactivity:</b><br>Nature of radiation from radioactive substances nuclear structure and nuclear properties. |
| <b>25<sup>th</sup> August</b>  | <b>SUNDAY</b>  |

|  |   |
|--|---|
| 26 <sup>th</sup> -31 <sup>st</sup> August      | <b>Nuclear Chemistry and Radioactivity:</b><br>Nuclear reaction, radioactive disintegration series, kinetics of radioactive disintegration.<br>Artificial transmutation of elements.  |
| 1 <sup>st</sup> September                      | <b>SUNDAY</b>   |
| 2 <sup>nd</sup> -7 <sup>th</sup> September     | <b>Nuclear Chemistry and Radioactivity:</b><br>Nuclear fission and nuclear fusion. Radio - carbondating , synthetic elements. Composition of nuclei: forces operating within the nucleus,nuclear stability and mass energy. |
| 08 <sup>th</sup> September                     | <b>SUNDAY</b>   |
| 9 <sup>th</sup> -14 <sup>th</sup> September    | <b>Nuclear Chemistry and Radioactivity:</b><br>Types of nuclear reaction. The compound nucleartheory, scintillation counters. Activation analysis. Isotopic dilution and radioactivetitration application.                  |
| 15 <sup>th</sup> September                     | <b>SUNDAY</b>   |
| 16 <sup>th</sup> -21 <sup>st</sup> September   | <b>Polymers Chemistry:</b><br>Polymerisation, classification of polymers, natural and synthetic polymers. Generalmethods of preparation. addition and condensation polymer's.   |
| 22 <sup>nd</sup> September                     | <b>SUNDAY</b>   |
| 23 <sup>th</sup> -28 <sup>th</sup> September   | <b>Polymers Chemistry:</b><br>Number average molecularweight, Weight average molecular weight.  |
| 29 <sup>th</sup> September                     | <b>SUNDAY</b>   |
| 30 <sup>th</sup> Sep -05 <sup>th</sup> October | <b>Polymers Chemistry:</b><br>Determination of molecular weight byosmotic, pressure method, viscosity method, light scattering method, kinetics ofcondensation polymerization.  |
| 06 <sup>th</sup> October                       | <b>SUNDAY</b>   |
| 07 <sup>th</sup> -12 <sup>th</sup> October     | <b>Polymers Chemistry:</b><br>kinetics of chain polymerisation, kinetics of cationic,anonic and condensation polymerisation. Copolymerisation.  |
| 13 <sup>th</sup> October                       | <b>SUNDAY</b>   |
| 14 <sup>th</sup> -19 <sup>th</sup> October     | <b>Physical properties and Molecular structure:</b><br>Optical activity, polarization, clausius- mossotti equation, orientation of dipoles in electric field.   |
| 20 <sup>th</sup> October                       | <b>SUNDAY</b>   |
| 21 <sup>st</sup> -26 <sup>th</sup> October     | <b>Physical properties and Molecular structure:</b><br>Dipole moment, induced dipole moment, measurement of dipole moment bytemperature methods and refractivity method.  |
| 27 <sup>th</sup> October                       | <b>SUNDAY</b>   |
| 4 <sup>th</sup> -9 <sup>th</sup> November      | <b>Physical properties and Molecular structure:</b><br>Dipolmoment and chemical constitution,magnetic properties.   |
| 10 <sup>th</sup> November                      | <b>SUNDAY</b>   |

|   |   |
|---|---|
| 11 <sup>th</sup> -16 <sup>th</sup><br>November      | <b>Physical properties and Molecular structure:</b><br><br>paramagnetic diamagnetic ferrodynamic. |
| 17 <sup>th</sup> November                           | <b>SUNDAY</b>   |
| 18 <sup>th</sup> November<br>onwards till<br>Exams. | Revision, Assignment, Test  |

| <b>Session: 2024-25</b>                                   |  |
|---|--|
| Name of teacher- Anil, Rinku, Nidhi Mann, Dr. Poonam Devi |  |
| Class- B.Sc. (Life Sciences/Physical Sciences)            |  |
| Subject- DSC Paper – I Fundamental Chemistry              |  |
| WEEKS   | SYLLABUS   |
| 22-7-2024 to<br>27-7-2024                                 | <b>Unit-I</b><br><b>Chemical Bonding and Molecular Structure</b><br>Ionic bond, lattice energy, Born-Haber cycle and its applications, Fajan's rules, hydration energy, bond moment, dipole moment and percentage ionic character.   |
| 29-7-2024 to 3-<br>8-2024                                 | Resonance and resonance energy: study of some inorganic and organic compounds. Molecular Orbital Approach: LCAO method, bonding and antibonding MOs and their characteristics for s-s, s-p and p-p combination of atomic orbitals, non- bonding combination of orbitals,   |
| 5-8-2024 to 10-<br>8-2024                                 | MO treatment of homonuclear diatomic molecules of 1st and 2nd periods (including idea of s-p mixing) and heteronuclear diatomic molecules such as O <sub>2</sub> <sup>-</sup> , O <sub>2</sub> <sup>2-</sup> , N <sub>2</sub> <sup>-</sup> , CO, NO <sup>+</sup> , CN <sup>-</sup> . Comparison of VB and MO approaches                |
| 12-8-2024 to<br>17-8-2024                                 | <b>Unit-II</b><br><b>p-Block Elements</b><br>Oxides – structures of oxides of N, P. Oxyacids – structure and relative acid strengths of oxyacids of nitrogen and phosphorus  |
| 20-8-2024 t 24-8-<br>2024                                 | . Structure of white, yellow and red phosphorus. Oxyacids of sulphur – structures and acidic strength, H <sub>2</sub> O <sub>2</sub> –structure, properties and uses. Basic properties of halogen, interhalogen compounds- types and properties, halogen-acids and oxyacids of chlorine – structure and comparison of acidic strength. |
| 27-8-2024 to 31-8-<br>2024                                | <b>Acids and Bases:</b> Brönsted–Lowry concept, conjugate acids and bases, relative strengths of acids and bases, effects of substituent and solvent, differentiating and levelling solvents   |
| 2-9-2024 to 7-9-<br>2024                                  | . Lewis acid-base concept, classification of Lewis acids and bases, Lux-Flood concept.   |

|                          |  |
|--------------------------|--|
| 9-9-2024 to 14-9-2024    | <p><b>Unit–III</b><br/> <b>Gaseous States</b><br/> Maxwell’s distribution of velocities and energies (derivation excluded), calculation of root mean square velocity, average velocity and most probable velocity.</p>   |
| 16-9-2024 to 21-9-2024   | Collision diameter, collision number, collision frequency and mean free path, deviation of real gases from ideal behaviour, derivation of Van der Waals Equation of state and its applications in the calculation of Boyle’s temperature (compression factor),   |
| 23-9-2024 to 28-9-2024   | explanation of behavior of real gases using Van der Waals equation.  |
| 30-9-2024 to 5-10-2024   | <b>Critical Phenomenon:</b> Critical temperature, critical pressure, critical volume and their determination. PV isotherms of real gases, continuity of states, isotherms of Van der Waals equation, relationship between critical constants and Van der Waals constants, compressibility factor. Law of corresponding |
| 7-10-2024 to 12-10-2024  | <p><b>Unit–IV</b><br/> <b>Basics of Organic Chemistry and Stereochemistry</b><br/> Electronic displacements and its applications, reaction intermediates and concept of aromaticity.</p>   |
| 14-10-2024 to 19-10-2024 | Concept of isomerism, types of isomerism, optical isomerism, optical activity, elements of symmetry, molecular chirality, enantiomers, stereogenic centre,   |
| 21-10-2024 to 26-10-2024 | properties of enantiomers, chiral and achiral molecules with two stereogenic centres, diastereomers, threo and erythro diastereomers   |
| 4-11-2024 to 9-11-2024   | , meso compounds, resolution of enantiomers, inversion, retention and racemization, relative and absolute configuration, sequence rules, R & S system of nomenclature.   |
| 11-11-2024 to 20-11-2024 | Unit Test  |
| 23-11-2024 to 20-12-2024 | <b>MDU examination</b>   |
| 21-12-2024 to 31-12-2024 | <b>Winter break</b>  |

**LESSON PLAN- B.Sc 3<sup>rd</sup> SEMESTER****Session: 2024-25**

Name of teacher- Dr. Jyoti, Lokesh  
Class- B.Sc. Pass Course (Medical and Non-medical)  
Subject- Inorganic Chemistry

| WEEKS                  | SYLLABUS  |
|------------------------|---|
| 22-7-2024 to 27-7-2024 | <b>Unit 1: Chemistry of Elements of 1st transition series:</b><br>Definition of transition elements.  |
| 29-7-2024 to 3-8-2024  | Position of lanthanides in the periodic table   |
| 5-8-2024 to 10-8-2024  | General characteristics & properties of 1st transition elements.  |
| 12-8-2024 to 17-8-2024 | Structures & properties of some compounds of transition elements<br>– TiO <sub>2</sub> , VOCl <sub>2</sub> , FeCl <sub>3</sub> .  |
| 20-8-2024 to 24-8-2024 | Structures & properties of some compounds of transition elements<br>– CuCl <sub>2</sub> and Ni(CO) <sub>4</sub>   |
| 27-8-2024 to 31-8-2024 | <b>Unit 2: Chemistry of Elements of 2nd &amp; 3rd transition series:</b> General characteristics and properties of the 2nd and 3rd transition elements                  |
| 2-9-2024 to 7-9-2024   | Comparison of properties of 3d elements with 4d & 5d elements with reference only to ionic radii, oxidation state.  |
| 9-9-2024 to 14-9-2024  | Comparison of properties of 3d elements with 4d & 5d elements with reference only to magnetic and Spectral properties and stereochemistry.<br><b>Test and Revision.</b> |
| 16-9-2024 to 21-9-2024 | <b>Unit 3: Coordination Compounds:</b> Werner's coordination theory,  |
| 23-9-2024 to 28-9-2024 | effective atomic number concept, chelates.  |
| 30-9-2024 to 5-10-2024 | nomenclature of coordination compounds, isomerism in coordination compounds.  |

|                          |   |
|--------------------------|---|
|                          |   |
| 7-10-2024 to 12-10-2024  | valence bond theory of transition metal complexes   |
| 14-10-2024 to 19-10-2024 | <b>Unit 4: Non-aqueous Solvents:</b> Physical properties of a solvent,                                |
| 21-10-2024 to 26-10-2024 | types of solvents and their general characteristics   |
| 4-11-2024 to 9-11-2024   | reactions in non-aqueous solvents with reference to liquid NH <sub>3</sub> and liquid SO <sub>2</sub> |
| 11-11-2024 to 20-11-2024 | <b>Test, Revision, Assignments, Viva.</b>   |
| 23-11-2024 to 20-12-2024 | <b>MDU examination</b>  |
| 21-12-2024 to 31-12-2024 | <b>Winter break</b>   |

| <b>LESSON PLAN- B.Sc 3<sup>rd</sup> SEMESTER</b>   |   | <b>Session: 2024-25</b> |
|--|---|-------------------------|
| Name of teacher- Kiran Bala, Dr. Ruman Rani<br>Class- B.Sc. Pass Course (Medical and Non-medical)<br>Subject- Physical Chemistry |   |                         |
| <b>WEEKS</b>   | <b>SYLLABUS</b>   |                         |
| 22-7-2024 to 27-7-2024   | <b>Unit 1: Thermodynamics-I</b><br>Definition of thermodynamic terms: system, surrounding etc.  |                         |
| 29-7-2024 to 3-8-2024  | Types of systems, intensive and extensive properties. State and path functions and their differentials. Thermodynamic process. Concept of heat and work.  |                         |
| 5-8-2024 to 10-8-2024  | Zeroth Law of thermodynamics, First law of thermodynamics: statement, definition of internal energy and enthalpy. Heat capacity, heat capacities at constant volume and pressure and their relationship |                         |

|                          |  |
|--------------------------|--|
|                          |  |
| 12-8-2024 to 17-8-2024   | Joule's law – Joule – Thomson coefficient for ideal gas and real gas: and inversion temperature.<br><b>Test and Revision.</b>  |
| 20-8-2024 to 24-8-2024   | <b>Unit 2: Thermodynamics-II</b> Calculation of $w, q, dU$ & $dH$ for the expansion of ideal gases under isothermal and adiabatic conditions for reversible process. |
| 27-8-2024 to 31-8-2024   | Temperature dependence of enthalpy, Kirchoffs equation   |
| 2-9-2024 to 7-9-2024     | Bond energies and applications of bond energies.   |
| 9-9-2024 to 14-9-2024    | <b>Unit 3: Chemical Equilibrium</b> Equilibrium constant and free energy, concept of chemical potential,   |
| 16-9-2024 to 21-9-2024   | Thermodynamic derivation of law of chemical equilibrium. Temperature dependence of equilibrium constant; Van't Hoff reaction isochore, Van't Hoff reaction isotherm. |
| 23-9-2024 to 28-9-2024   | Le-Chatetier's principle and its applications Clapeyron equation and Clausius – Clapeyron equation its applications.   |
| 30-9-2024 to 5-10-2024   | <b>Unit 4: Distribution Law</b> Nernst distribution law – its thermodynamic derivation,  |
| 7-10-2024 to 12-10-2024  | Modification of distribution law when solute undergoes dissociation, association and chemical combination.   |
| 14-10-2024 to 19-10-2024 | Applications of distribution law: (i) Determination of degree of hydrolysis and hydrolysis constant of aniline hydrochloride.  |
| 21-10-2024 to 26-10-2024 | (ii) Determination of equilibrium constant of potassium tri-iodide complex and process of extraction.  |
| 4-11-2024 to 9-11-2024   | <b>Test, Revision.</b>   |

|                          |                           |
|--------------------------|---------------------------|
| 11-11-2024 to 20-11-2024 | <b>Assignments, Viva.</b> |
| 23-11-2024 to 20-12-2024 | <b>MDU examination</b>    |
| 21-12-2024 to 31-12-2024 | <b>Winter break</b>       |

| <b>LESSON PLAN- B.Sc 3<sup>rd</sup> SEMESTER</b> |   | <b>Session: 2024-25</b> |
|--|---|-------------------------|
| Name of teacher- Dr. Poonam,, Manoj              |   |                         |
| Subject- organic Chemistry                       |   |                         |
| <b>WEEKS</b>                                     | <b>SYLLABUS</b>   |                         |
| 22-7-2024 to 27-7-2024                           | <b>Unit 1: Alcohols</b><br><br>Monohydric alcohols nomenclature, methods of formation by reduction of aldehydes, ketone.                                    |                         |
| 29-7-2024 to 3-8-2024                            | Methods of formation by reduction of carboxylic acids and esters. Hydrogen bonding. Acidic nature.  |                         |
| 5-8-2024 to 10-8-2024                            | Reactions of alcohols. Dihydric alcohols — nomenclature, methods of formation, chemical reactions of vicinal glycols  |                         |
| 12-8-2024 to 17-8-2024                           | <b>Unit 2 Epoxides</b><br><br>Synthesis of epoxides. Acid and base-catalyzed ring opening of epoxides, orientation of epoxide ring opening                  |                         |
| 20-8-2024 to 24-8-2024                           | Reactions of Grignard and organolithium reagents with epoxides  |                         |
| 27-8-2024 to 31-8-2024                           | <b>Unit 3: Phenols</b><br><br>Nomenclature, structure and bonding. Preparation of phenols, physical properties and acidic character.                        |                         |
| 2-9-2024 to 7-9-2024                             | Comparative acidic strengths of alcohols and phenols, resonance stabilization of phenoxide ion. Reactions of phenols — electrophilic aromatic substitution, |                         |

|                          |   |
|--------------------------|---|
|                          |   |
| 9-9-2024 to 14-9-2024    | Mechanisms of Fries rearrangement, Claisen rearrangement, Reimer-Tiemann reaction, Kolbe's reaction. Schotten and Baumann reactions   |
| 16-9-2024 to 21-9-2024   | <b>Unit 4: Ultraviolet (UV) absorption spectroscopy</b><br><br>Absorption laws (Beer-Lambert law), molar absorptivity, presentation and analysis of UV spectra<br><br>Types of electronic transitions, effect of conjugation. Concept of chromophore and auxochrome   |
| 23-9-2024 to 28-9-2024   | .Bathochromic, hypsochromic, hyperchromic and hypochromic shifts. UV spectra of conjugated dienes and enones<br><br>Woodward-Fieser rules, calculation of max of simple conjugated dienes and $\alpha,\beta$ -unsaturated ketones.<br><br>Applications of UV Spectroscopy in structure elucidation of simple organic compounds. |
| 30-9-2024 to 5-10-2024   | <b>Unit 5: Carboxylic Acids &amp; Acid Derivatives</b><br><br>Nomenclature of Carboxylic acids, structure and bonding, physical properties, acidity<br><br>of carboxylic acids,   |
| 7-10-2024 to 12-10-2024  | Effects of substituents on acid strength. Preparation of carboxylic acids. Reactions of carboxylic acids.   |
| 14-10-2024 to 19-10-2024 | Hell-Volhard-Zelinsky reaction. Reduction of carboxylic acids. Mechanism of decarboxylation. Structure, nomenclature and preparation of acid chlorides,   |
| 21-10-2024 to 26-10-2024 | Structures, nomenclature and preparation of esters, amides and acid anhydrides.<br><br>Relative stability of acyl derivatives. Physical properties,   |

|                          |  |
|--------------------------|--|
|                          | interconversion of acid derivatives by nucleophilic acyl substitution. |
| 4-11-2024 to 9-11-2024   | Mechanisms of esterification and hydrolysis (acidic and basic)         |
| 11-11-2024 to 20-11-2024 | Assignments, Viva, Test, Revision                                      |
| 23-11-2024 to 20-12-2024 | MDU examination  |
| 21-12-2024 to 31-12-2024 | Winter break   |

| <b>LESSON PLAN- B.Sc5th SEMESTER</b> <b>Session: 2024-25</b> |  |
|--|--|
| Name of teacher- Bhupender Singh, Rekha Gautam               |  |
| Subject- Inorganic Chemistry                                 |  |
| WEEKS  | SYLLABUS   |
| 22-7-2024 to 27-7-2024                                       | Metal-ligand Bonding in Transition Metal Complexes Limitations of valence bond theory,             |
| 29-7-2024 to 3-8-2024  | an elementary idea of crystal-field theory, crystal field splitting in octahedral                  |
| 5-8-2024 to 10-8-2024  | crystal field split tetrahedral and square planar complexes,                                       |
| 12-8-2024 to 17-8-2024                                       | factors affecting the crystal-field parameters   |
| 20-8-2024 to 24-8-2024                                       | Thermodynamic and Kinetic Aspects of Metal Complexes   |
| 27-8-2024 to 31-8-2024                                       | A brief outline of thermodynamic stability of metal complexes and factors affecting the stability, |
| 2-9-2024 to 7-9-2024   | substitution reactions of square planar complexes of Pt(II)  |
| 9-9-2024 to 14-9-2024  | Magnetic Properties of Transition Metal Complexes Types of magnetic behaviour                      |

|                          |  |
|--------------------------|--|
| 16-9-2024 to 21-9-2024   | methods of determining magnetic susceptibility, spin-only formula.   |
| 23-9-2024 to 28-9-2024   | L-S coupling, correlation of $s$ and $eff$ values, orbital contribution to magnetic moments, application of magnetic moment data for 3d metal complexes. |
| 30-9-2024 to 5-10-2024   | Electron Spectra of Transition Metal Complexes Types of electronic transitions,  |
| 7-10-2024 to 12-10-2024  | selection rules for d-d transitions, spectroscopic ground states, spectrochemical series.  |
| 14-10-2024 to 19-10-2024 | Orgel-energy level diagram for $d1$ and $d9$ states,   |
| 21-10-2024 to 26-10-2024 | discussion of the electronic spectrum of $[Ti(H_2O)_6]^{3+}$ complex ion   |
| 4-11-2024 to 9-11-2024   | Revision of Electron Spectra , Doubts  |
| 11-11-2024 to 20-11-2024 | Assignments, Viva, Test, Revision  |
| 23-11-2024 to 20-12-2024 | MDU examination  |
| 21-12-2024 to 31-12-2024 | Winter break   |

| <b>LESSON PLAN- B.Sc 5th SEMESTER</b>   |   | <b>Session: 2024-25</b> |
|---|---|-------------------------|
| Name of teacher- Sandeep Kumar, Praveen |   |                         |
| Subject- Physical Chemistry             |   |                         |
| WEEKS                                   | SYLLABUS  |                         |
| 22-7-2024 to 27-7-2024                  | Quantum Mechanic s-Black-body radiation, Plank’s radiation law, photoelectric effect  |                         |
| 29-7-2024 to 3-8-2024                   | heat capacity of solids, Compton effect, wave function and its significance of Postulates of quantum mechanics ,.   |                         |
| 5-8-2024 to 10-8-2024                   | quantum mechanical operator, commutation relations, Hamiltonian operator, Hermitian operator, average value of square of Hermitian as a positive quantity,  |                         |
| 12-8-2024 to 17-8-2024                  | Role of operators in quantum mechanics, To show quantum mechanically that position and momentum cannot be predicated simultaneously,  |                         |
| 20-8-2024 t 24-8-2024                   | Determination of wave function & energy of a particle in one dimensional box, Pictorial representation and its significance   |                         |
| 27-8-2024 to 31-8-2024                  | Physical Properties and Molecular Structure Optical activity, polarization – (clausius – Mossotti equation). Orientation of dipoles in an electric field, dipole moment,                                    |                         |
| 2-9-2024 to 7-9-2024                    | , measurement of dipole moment-temperature method and refractivity method, dipole moment and  |                         |
| 9-9-2024 to 14-9-2024                   | structure of molecules, Magnetic permeability, magnetic susceptibility and its determination. Application of magnetic susceptibility, magnetic properties – Para magnetism, diamagnetism and ferromagnetic. |                         |
| 16-9-2024 to 21-9-2024                  | Spectroscopy-I Introduction: Electromagnetic radiation, regions of spectrum, basic features of spectroscopy, statement of Born Oppenheimer approximation, Degrees of freedom..                              |                         |

|                          |  |
|--------------------------|--|
|                          |  |
| 23-9-2024 to 28-9-2024   | Rotational Spectrum Diatomic molecules. Energy levels of rigid rotator (semi-classical principles), selection rules,   |
| 30-9-2024 to 5-10-2024   | spectral intensity distribution using population distribution (Maxwell-Boltzmann distribution), determination of bond length, qualitative description of non-rigid rotor, isotope effect.. |
| 7-10-2024 to 12-10-2024  | Vibrational spectrum Infrared spectrum: Energy levels of simple harmonic oscillator,   |
| 14-10-2024 to 19-10-2024 | selection rules, pure vibrational spectrum, intensity, determination of force constant and qualitative relation of force constant and bond energies,                                       |
| 21-10-2024 to 26-10-2024 | effects of anharmonic motion and isotopic effect on the spectra., idea of vibrational frequencies of different functional groups.  |
| 4-11-2024 to 9-11-2024   | Raman Spectrum: Concept of polarizability, pure rotational and pure vibrational Raman spectra of diatomic molecules, selection rules, Quantum theory of Raman spectra                      |
| 11-11-2024 to 20-11-2024 | Assignments, Viva, Test, Revision  |
| 23-11-2024 to 20-12-2024 | MDU examination  |
| 21-12-2024 to 31-12-2024 | Winter break   |

**LESSON PLAN- B.Sc 5th SEMESTER****Session: 2024-25**

Name of teacher- Reena, Neeraj

Subject- Organic Chemistry

| WEEKS                         | SYLLABUS  |
|-------------------------------|---|
| 22-7-2024<br>to 27-7-<br>2024 | NMR Spectroscopy-I Principle of nuclear magnetic resonance, the PMR spectrum, number of signals, peak areas,.   |
| 29-7-2024<br>to 3-8-<br>2024  | equivalent and nonequivalent protons positions of signals and chemical shift, shielding and deshielding of protons,   |
| 5-8-2024<br>to 10-8-<br>2024  | proton counting, splitting of signals and coupling constants, magnetic equivalence of protons   |
| 12-8-2024<br>to 17-8-<br>2024 | NMR Spectroscopy-II Discuss ion of PMR spectra of the molecules: ethyl bromide, npropyl bromide, isopropyl bromide,   |
| 20-8-2024 t<br>24-8-2024      | 1,1-dibromoethane, 1,1,2-tribromoethane, ethanol, acetaldehyde, ethyl acetate, toluene,   |
| 27-8-2024 to<br>31-8-2024     | benzaldehyde and acetophenone. Simple problems on PMR spectroscopy for structure determination of organic compounds   |
| 2-9-2024 to 7-<br>9-2024      | Carbohydrates-I Classification and nomenclature. Monosaccharides, mechanism of osazone formation, interconversion of glucose and fructose, chain lengthening and chain shortening of aldoses. |

|                             |   |
|-----------------------------|---|
| 9-9-2024 to<br>14-9-2024    | Configuration of monosaccharides. Erythro and threo diastereomers. Conversion of glucose into mannose. Formation of glycosides,.        |
| 16-9-2024 to<br>21-9-2024   | ethers and esters. Determination of ring size of glucose and fructose. Open chain and cyclic structure of D(+)-glucose & D(-) fructose. |
| 23-9-2024 to<br>28-9-2024   | Mechanism of mutarotation. Structures of ribose and deoxyribose   |
| 30-9-2024 to<br>5-10-2024   | 1. Carbohydrates-II An introduction to disaccharides (maltose, sucrose and lactose) and.  |
| 7-10-2024 to<br>12-10-2024  | polysaccharides (starch and cellulose) without involving structure determination  |
| 14-10-2024 to<br>19-10-2024 | 2. Organometallic Compounds Organmagnesium compounds: the Grignard reagents-formation   |
| 21-10-2024 to<br>26-10-2024 | , structure and chemical reactions. Organozinc compounds: formation and   |
| 4-11-2024 to<br>9-11-2024   | chemical reactions. Organolithium compounds: formation and chemical reactions.  |
| 11-11-2024 to<br>20-11-2024 | Assignments, Viva, Test, Revision   |
| 23-11-2024 to<br>20-12-2024 | MDU examination   |
| 21-12-2024 to<br>31-12-2024 | Winter break  |